

EE 213, Microscopic Nanocharacterization of Materials
Lecture 10.
other micro-characterization using ion beams

Class website: <https://ee213-winter16-01.courses.soe.ucsc.edu>

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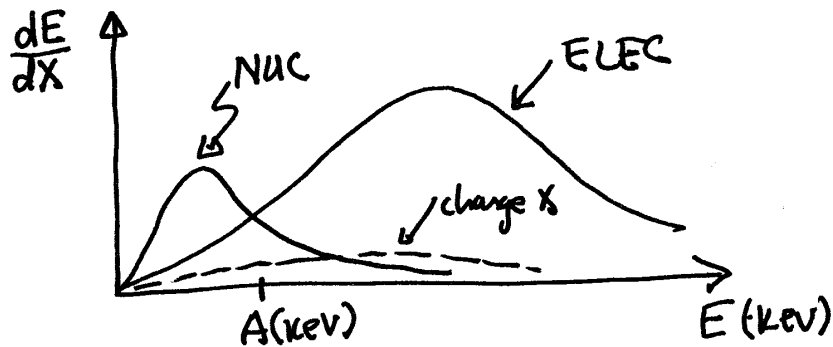
ION INTERACTIONS

three main components

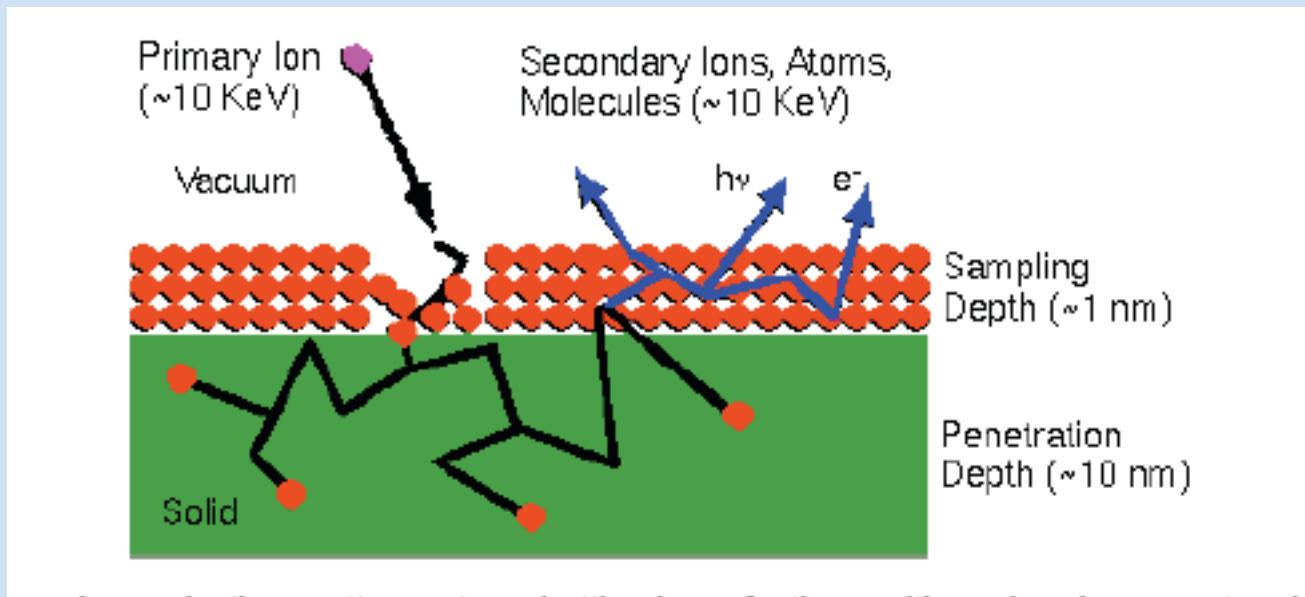
1. "nuclear" energy loss, dominates at low energies
when $E < A$ keV
at. wt. inc. ions

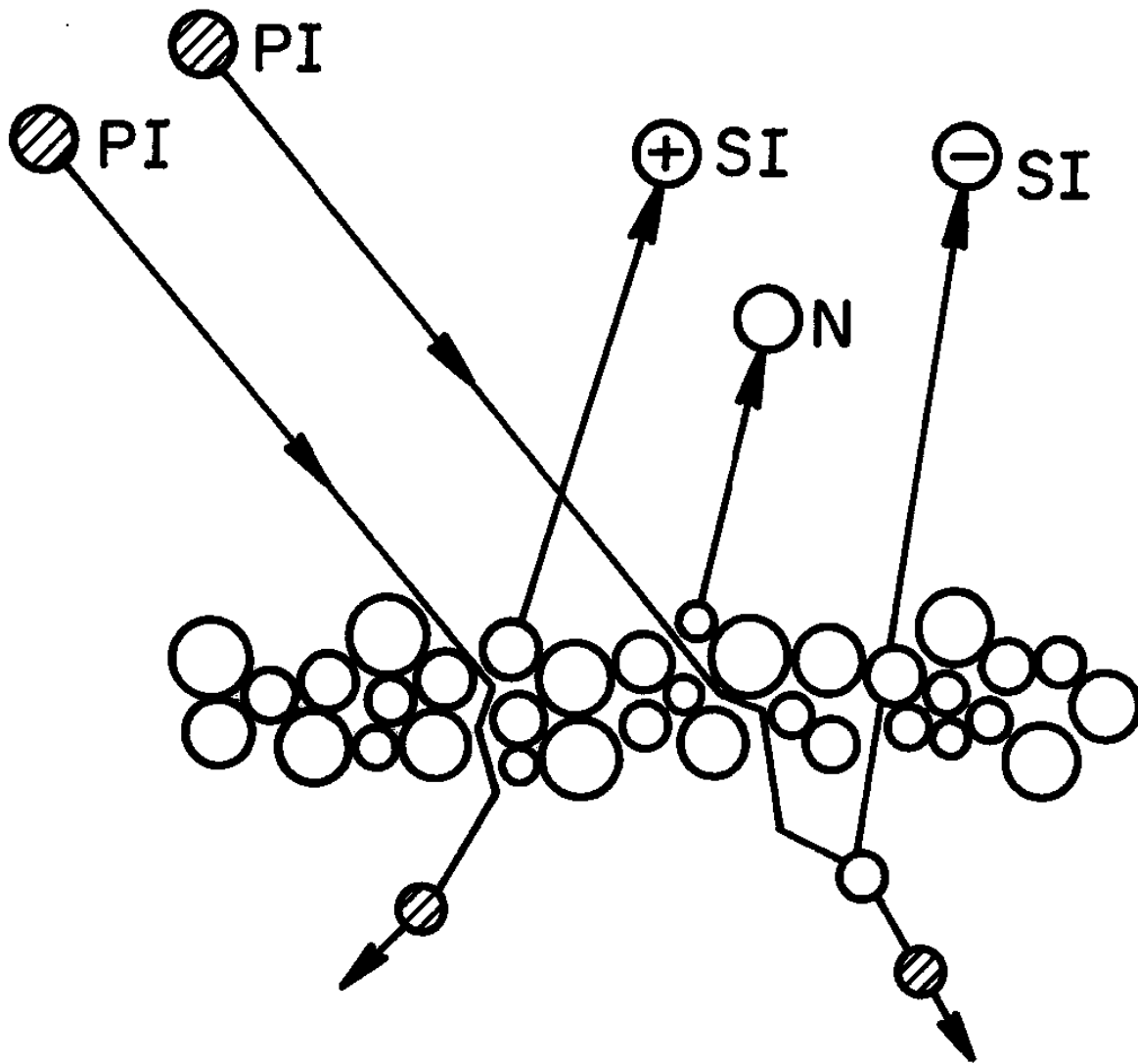
2. electronic energy loss
interactions with atomic electrons
(like "inelastic" electron scattering)

3. charge exchange
 $\lesssim 10\%$ of ~~the~~ total



Sputtering

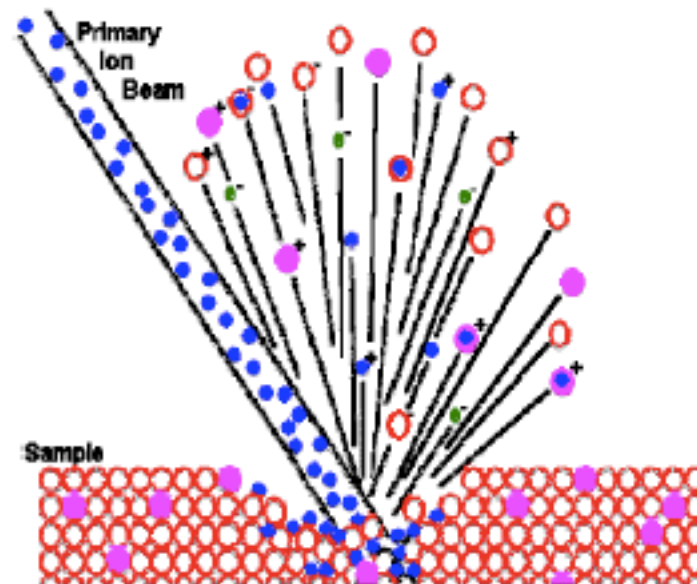




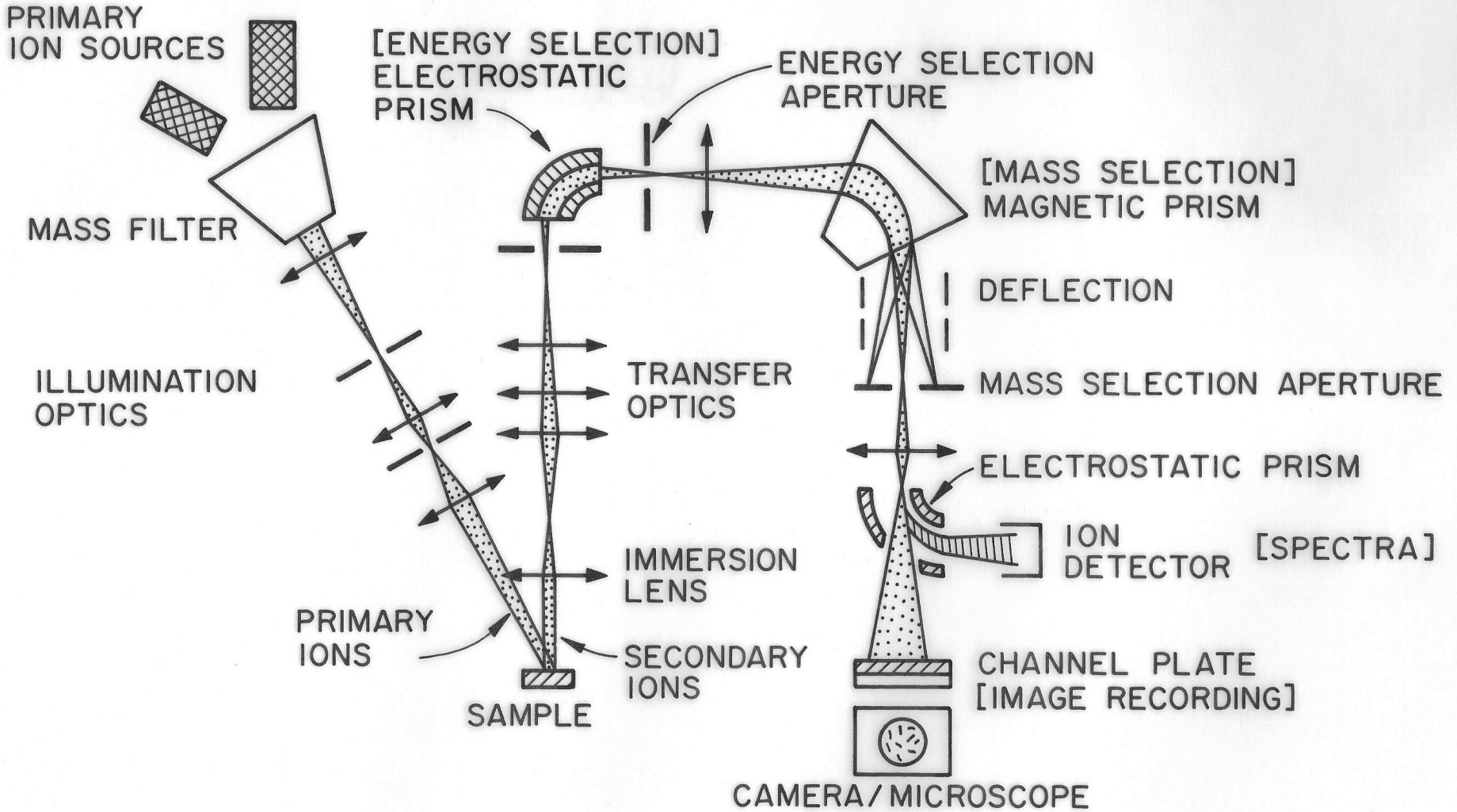
DEPTH OF EMISSION

Primary Ion Beam

- › Interaction with the surface known as sputtering
 - Amorphization of the surface
 - Implant Atoms
 - Secondary Particles are Ejected



<http://ppsm.tripod.com/SEM/Theory.htm>

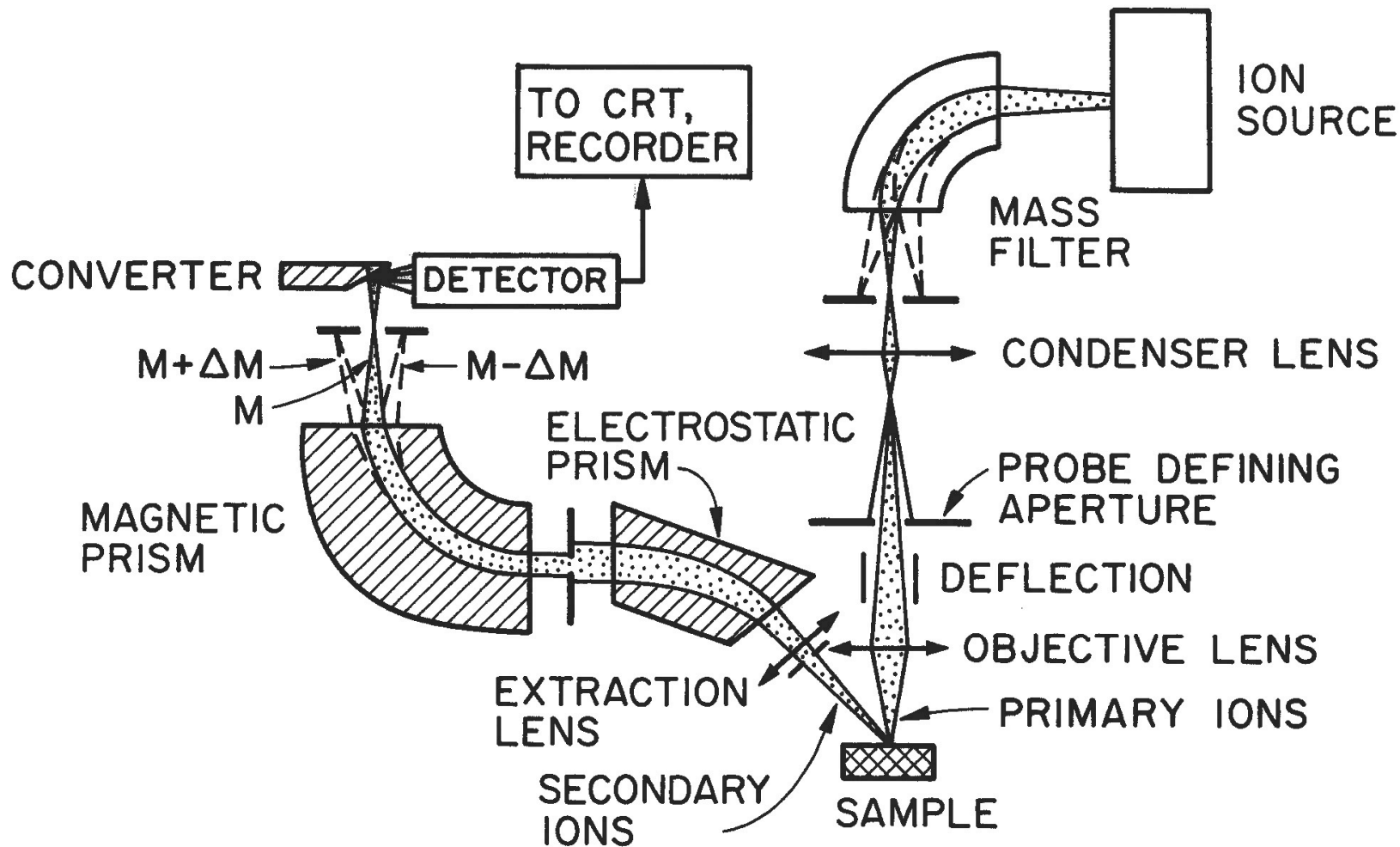


SECONDARY ION MICROSCOPE

Global Image

R. Castaing and G. Slodzian. *J. Microscopy*.1. (1962).395-410. (in French)

G. Slodzian. *Surface Science*.48.(1961)

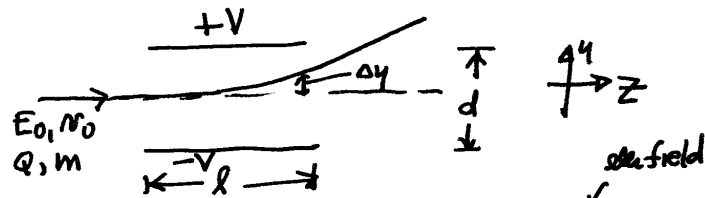


ION MICROPROBE

Scanning image

energy selection by electrostatic field /

genl: || plates



$$m\dot{y} = qE_y \rightarrow \begin{cases} \dot{y}_0 = 0, \dot{z}_0 = v_0 \\ y_0 = 0, z_0 = 0 \end{cases} \rightarrow \begin{cases} y = \frac{q}{2m} E_y t^2 \\ z = v_0 t \end{cases}$$

$$\therefore y = \frac{q}{2m} E_y \left(\frac{z_0}{v_0}\right)^2 \text{ but } E_0 = \frac{1}{2} m v_0^2$$

$$2E_0 = m v_0^2$$

$$\therefore y = q E_y z_0^2 / 4E_0 \quad E_y = \Delta V / d$$

$$\frac{E_0}{q}$$

|| defl $\propto \frac{1}{E_0}$ || an energy deflector

for spherical plates \Rightarrow balance $\vec{F} = Q\vec{E}$
with centrip force

radius E_0/q dependent || get $\frac{2E_0}{r} = Q E_0$ || smaller E_0 get bent more ||

$$r = \frac{2E_0}{QE_0}$$

diff than mag sector w/oo. $BR = \frac{mv}{Q}$ || momentum selection / or M/Q ratio if energy must

in a probe (a "scanning" technique.
 (in principle, simpler system / optics)

advantage: 1. can collect secondary electron image as well
 ≠

NOTE // ≠ different erosion process.

in all ~~the~~ SIMS ^{topographic} microscopy, structure of sample surface
 can affect results.

doc
 camera

- shadowing
- redeposition

what are detection limits with SIMS?

- it varies with atoms and matrix
- it varies whether + or - ions

show
 table //

example: $\text{in Si} / \text{Cu} \rightarrow \text{lim} = 5 \times 10^{13} / \text{cm}^3$
 with O_2^+
 $\text{lim} = 2 \times 10^{15} / \text{cm}^3$
 with Cs^-

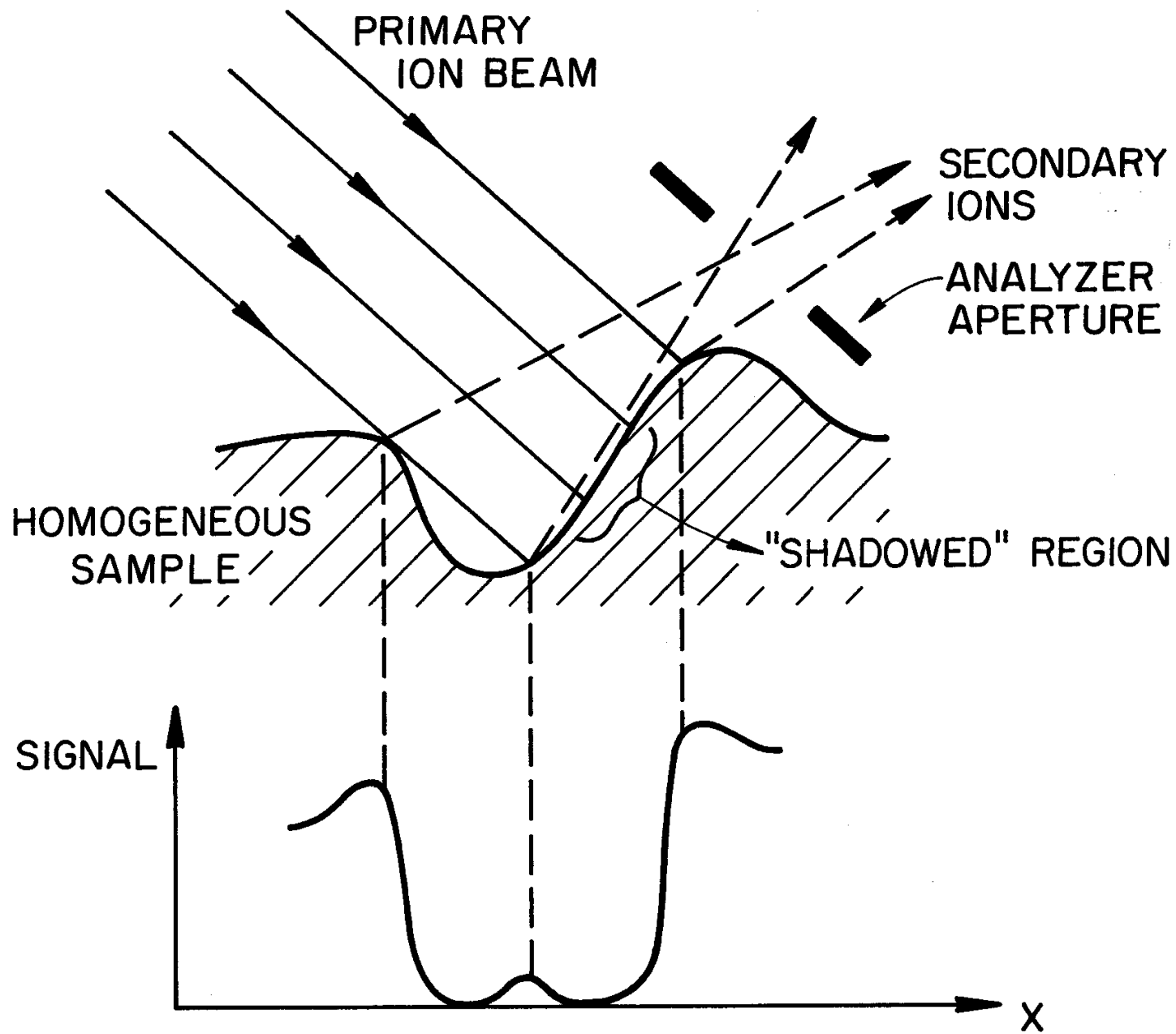
$\text{Si} // \rho = 2.7 \text{ gm/cm}^3$
 $A = 28 \text{ amu}$
 $N \cong 10^{-1} \times 6 \times 10^{23} \text{ atoms/cm}^3$
 $\cong 6 \times 10^{22} / \text{cm}^3 //$

$\text{in Si} / \text{Cu} \rightarrow \text{lim} = 2 \times 10^{14} / \text{cm}^3$
 with O_2^+
 $\text{lim} = 2 \times 10^{15} / \text{cm}^3$
 with Cs^-

NOTE / det limits are $10^{-6} - 10^{-7}$ atom % or better -
 almost a trace technique

Shadowing/Redeposition in SIMS

Rudenhauer and Steiger, Ultramicroscopy. 24 (2-3).1988.p.115-123



Detection Limits in SiO₂

O ₂ Primary Ion Beam Positive Ions		Cs Primary Ion Beam Negative Ions	
Element	DL (atoms/cm ³)	Element	DL (atoms/cm ³)
Li	1E+13	H	1E+18
B	5E+13	C	2E+17
N	2E+17	N	3E+16 – 2E+17
Na	2E+13	P	1E+15
Mg	2E+13	S	1E+16
Al	1E+14	Cl	1E+16
K	1E+13	Cr	2E+16
Ca	5E+13	Fe	3E+16
Ti	5E+13	Ni	3E+15
Cr	2E+13	Cu	2E+15
Mn	1E+14	As	2E+15
Co	2E+14	Ge	2E+15
Fe	2E+14 – 5E+14	Au	1E+15
Ni	1E+15	-	-
Cu	5E+14	-	-
Zn	3E+15	-	-
As	2E+16	-	-
Mo	5E+14	-	-
In	2E+14	-	-
Ta	5E+15	-	-
W	5E+15	-	-

Note: For dynamic sector O₂ Primary Ion Beam sputtering, the maximum oxide layer thickness should be less than 1.5µm for proper charging compensation.
Higher detection limit for BPSG

Discussion

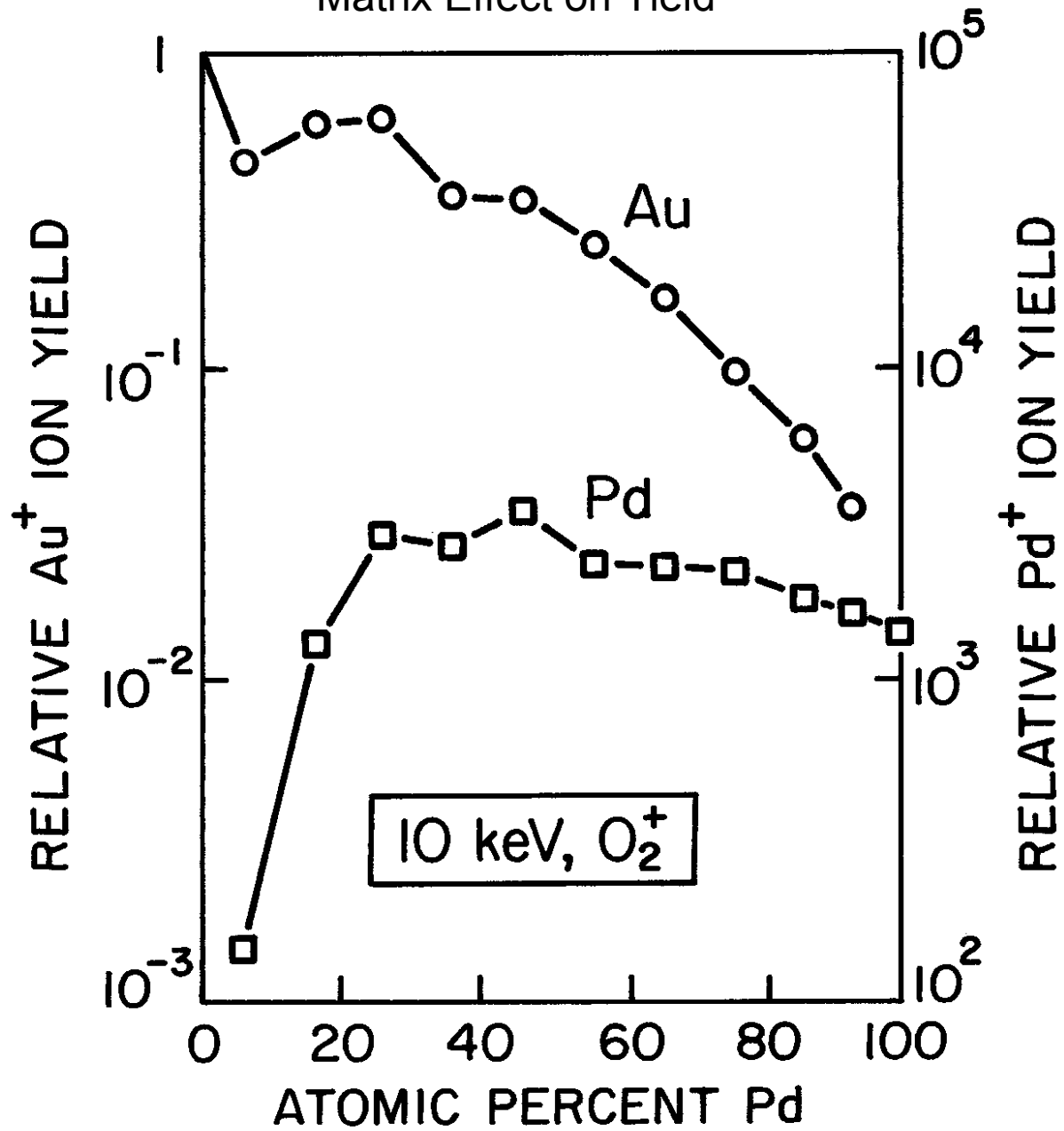
SIMS is a powerful analytical technique which allows detection of all elements from H to U with excellent sensitivity. The table provides a list of typical detection limits for impurities in Si and SiO₂ matrices. These detection levels are for normal depth profiling conditions of blanket wafers. Detection levels for device samples depends on the size of the available analysis area.

Detection Limits in Si

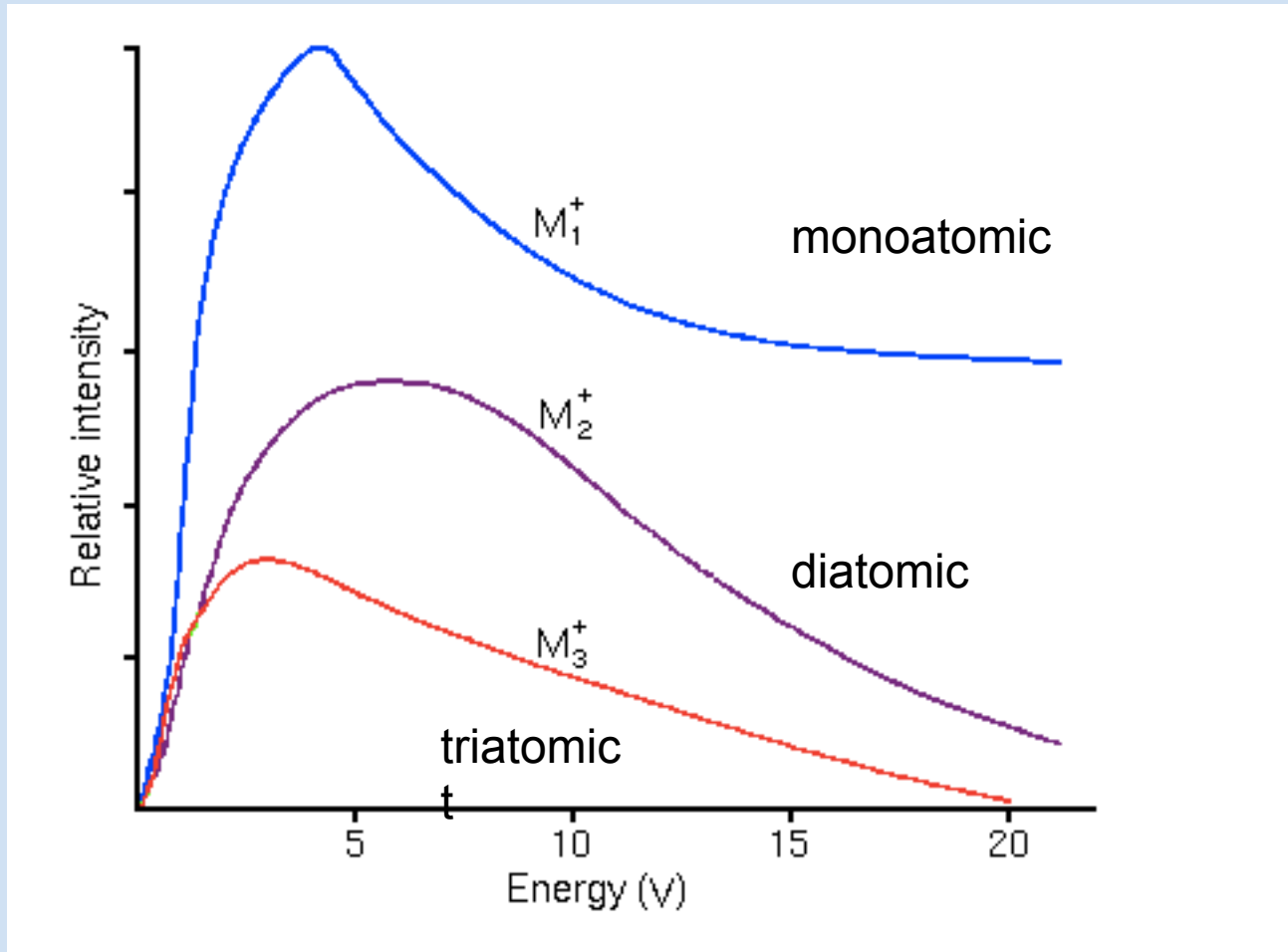
O ₂ Primary Ion Beam Positive Ions		Cs ⁺ Primary Ion Beam Negative Ions		Cs ⁺ Primary Ion Beam Positive Ions (MOCs)	
Element	DL (atoms/cm ³)	Element	DL (atoms/cm ³)	Element	DL (atoms/cm ³)
He	5E+17	H	1E+17	Ar	1E+17*
Li	5E+12	B	1E+15	-	-
B	2E+13	C	1E+16	-	-
Na	5E+12	N	1E+15	-	-
Mg	5E+12	O	5E+16	-	-
Al	2E+13	F	5E+15	-	-
K	5E+12	P	1E+14	-	-
Ca	1E+13	S	1E+15	-	-
Ti	1E+13	Cl	5E+15	-	-
Cr	2E+13	Cu	2E+15	-	-
Mn	2E+13	As	5E+13 -- 2E+15	-	-
Fe	5E+13 -- 2E+15	Ge	2E+14	-	-
Ni	5E+14	Sb	1E+14 -- 2E+15	-	-
Cu	2E+14	Au	5E+13	-	-
Zn	5E+15	-	-	-	-
As	5E+16	-	-	-	-
Mo	1E+14	-	-	-	-
In	5E+13	-	-	-	-
Ta	5E+14	-	-	-	-
W	2E+14	-	-	-	-

* Assuming Ca level is below 1E+15 atoms

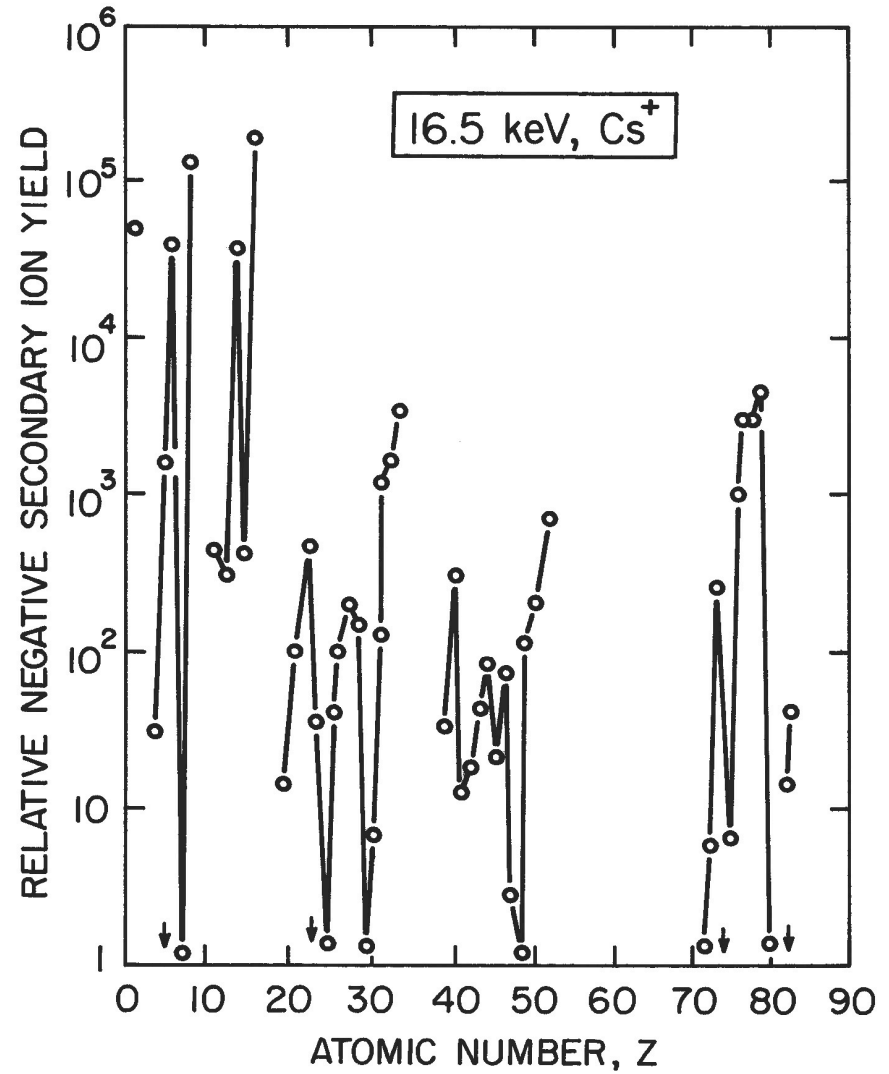
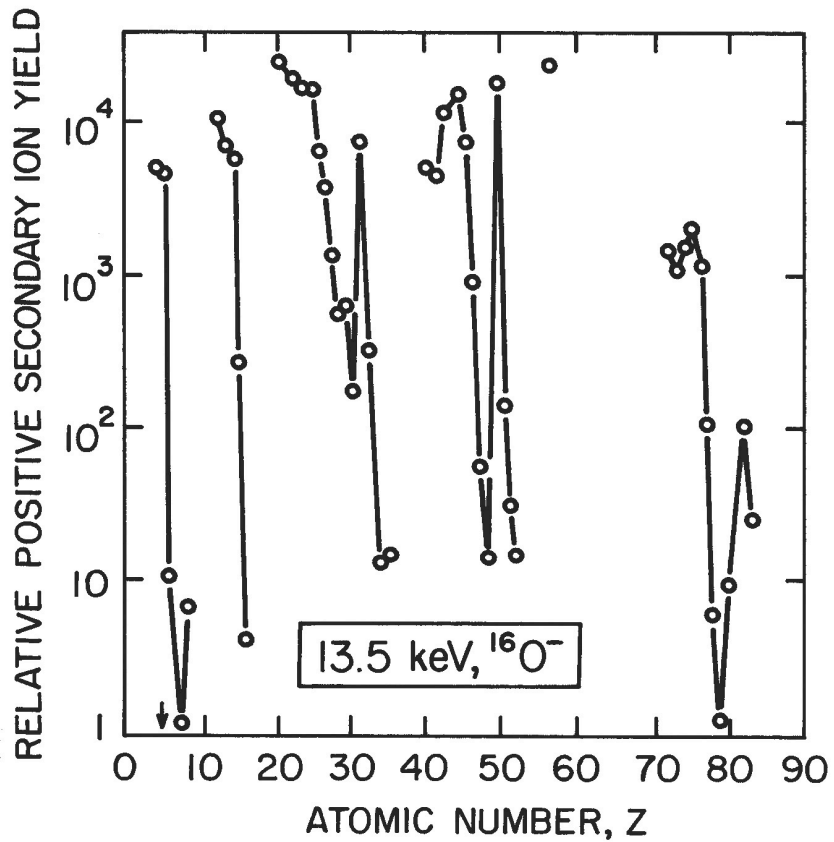
Matrix Effect on Yield

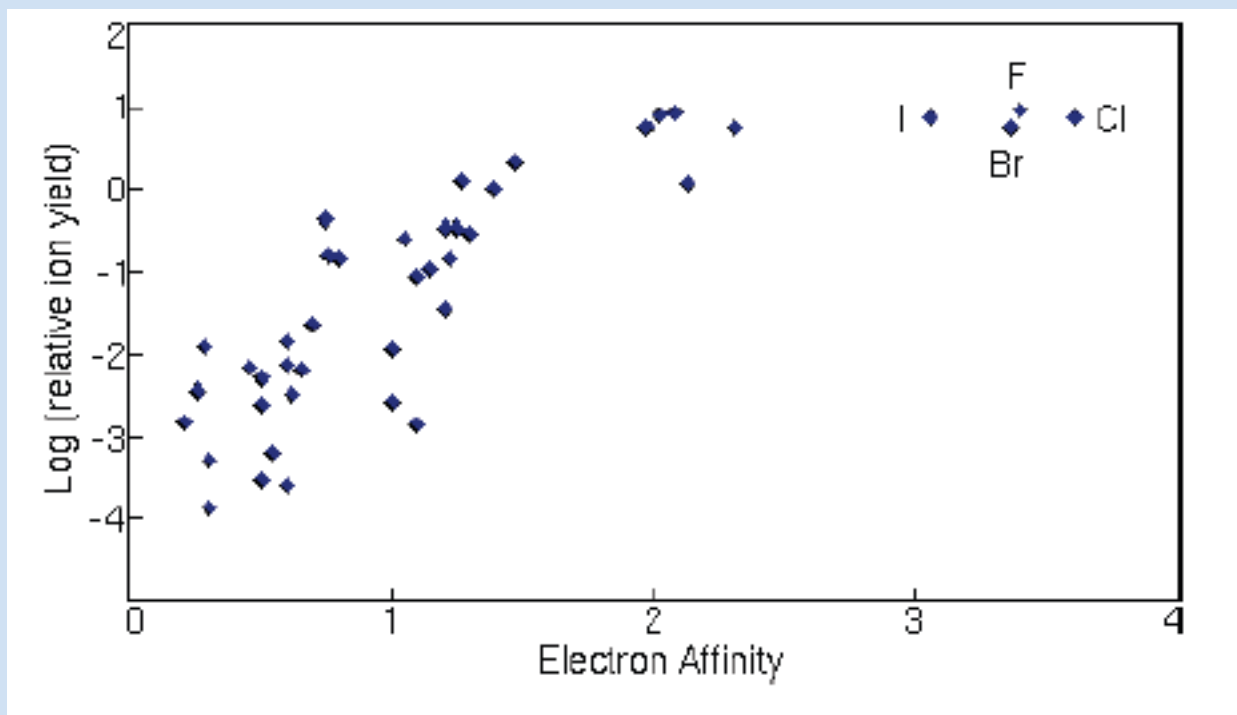


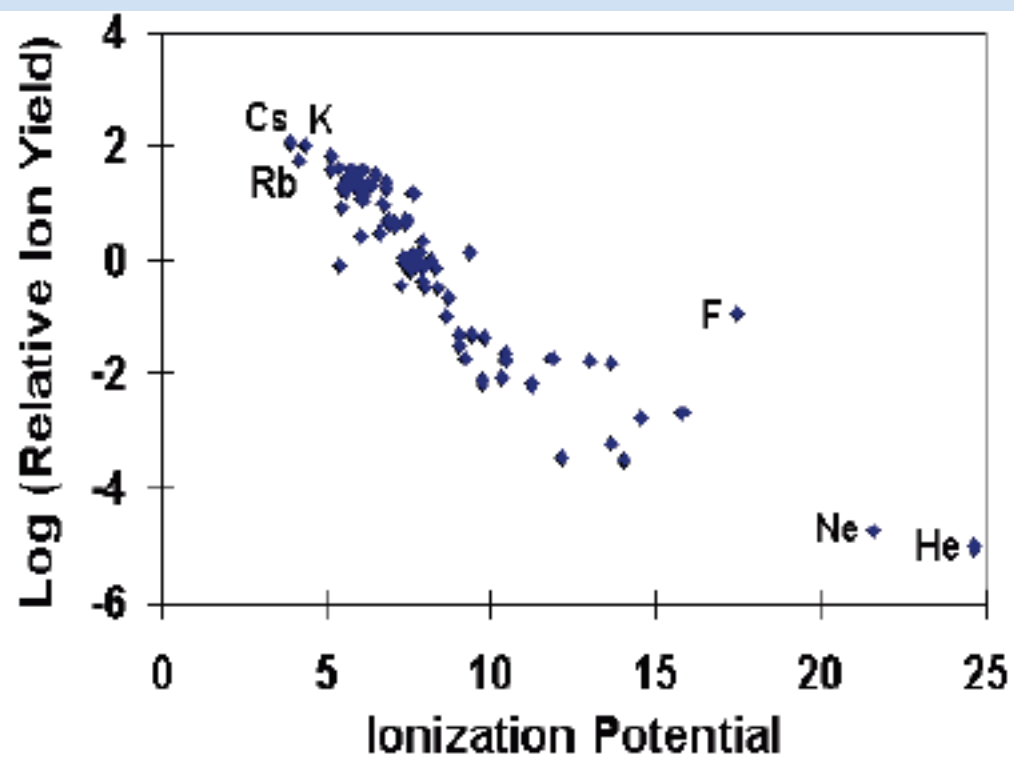
Secondary ion energy distributions



Secondary Ion Yield Depends on Lots of Things







how do you get depth profile?

measure ion emitted species as a function of time
and "calibrate" the sputtering process.

IR, time sputtering \Rightarrow avoid amount of material removed /
non-trivial since sputtering yield
varies depending upon material

- NOTE // depth resolution //

the instrumentation / - 2 types // scanning probe
2 global

ion Microscope / Cathodoluminescence ~ 60's //

what is needed?

1. ion source - needs to be mass filtered.
2. illumination optics
to focus into sample.
3. after sample: immersion optics
to image the secondary ions.

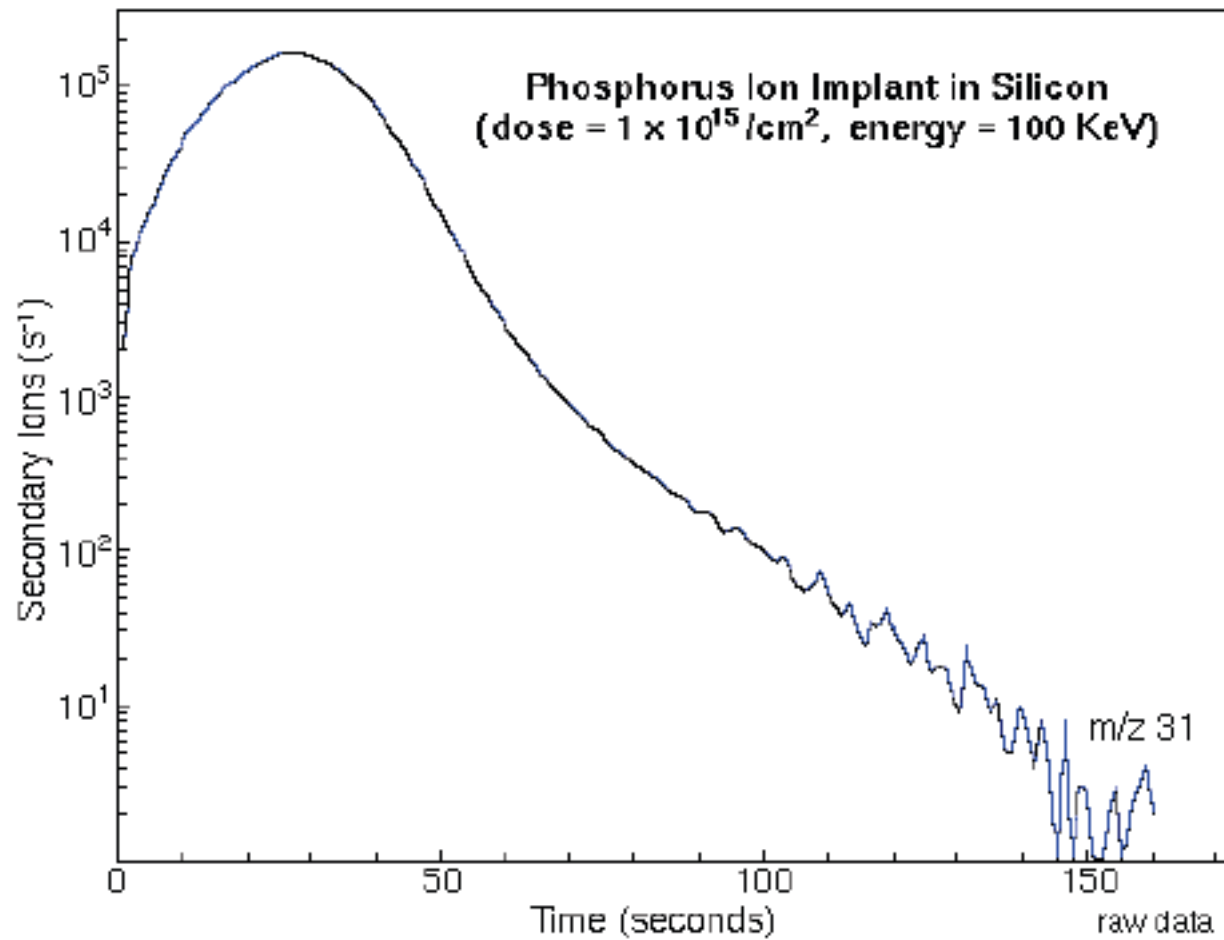
NOTE // can also image secondary elec!

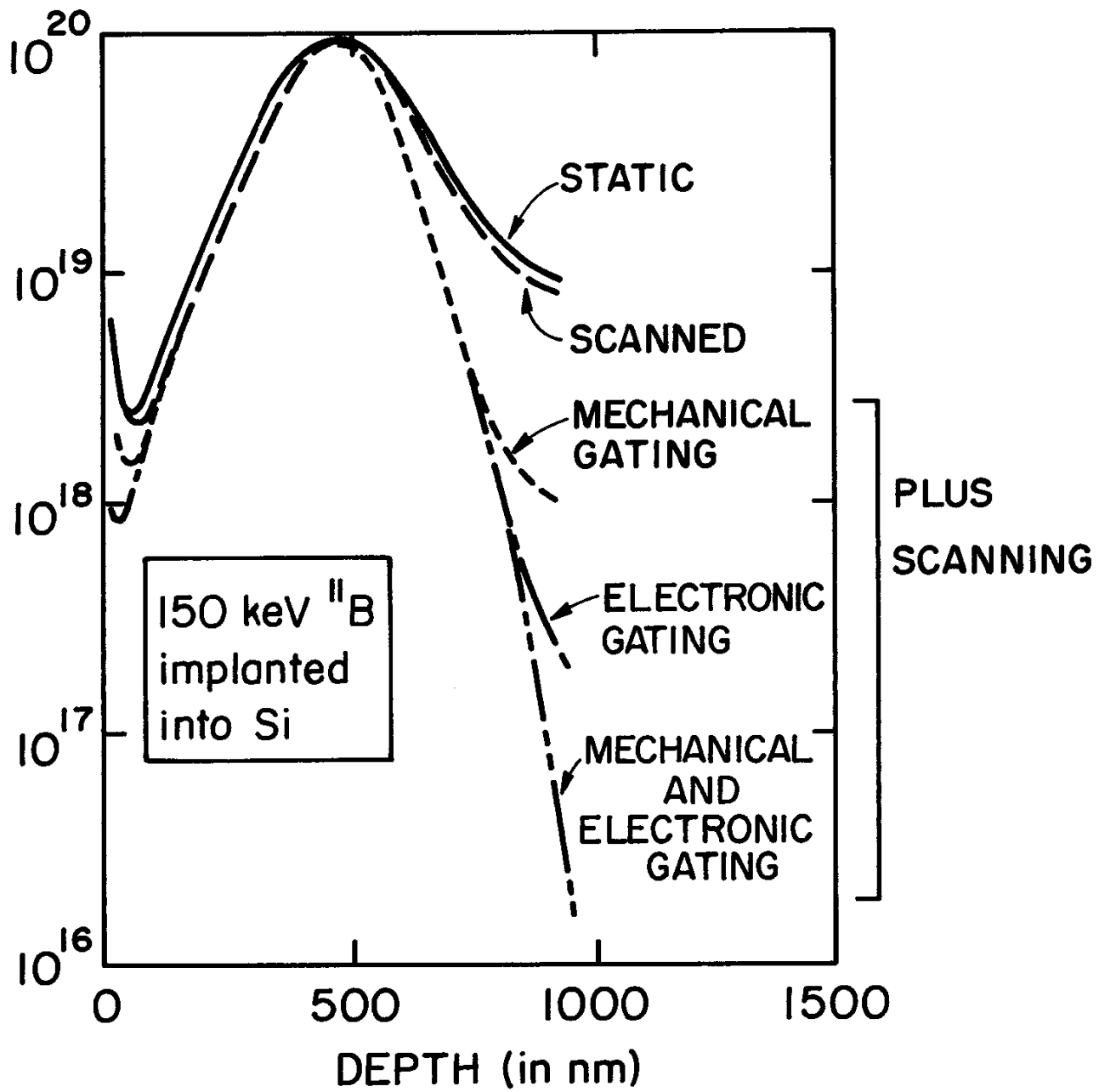
4. energy selection
(ion ions off with different energies)
- electrostatic prism
(an "energy separator" not a momentum one!)

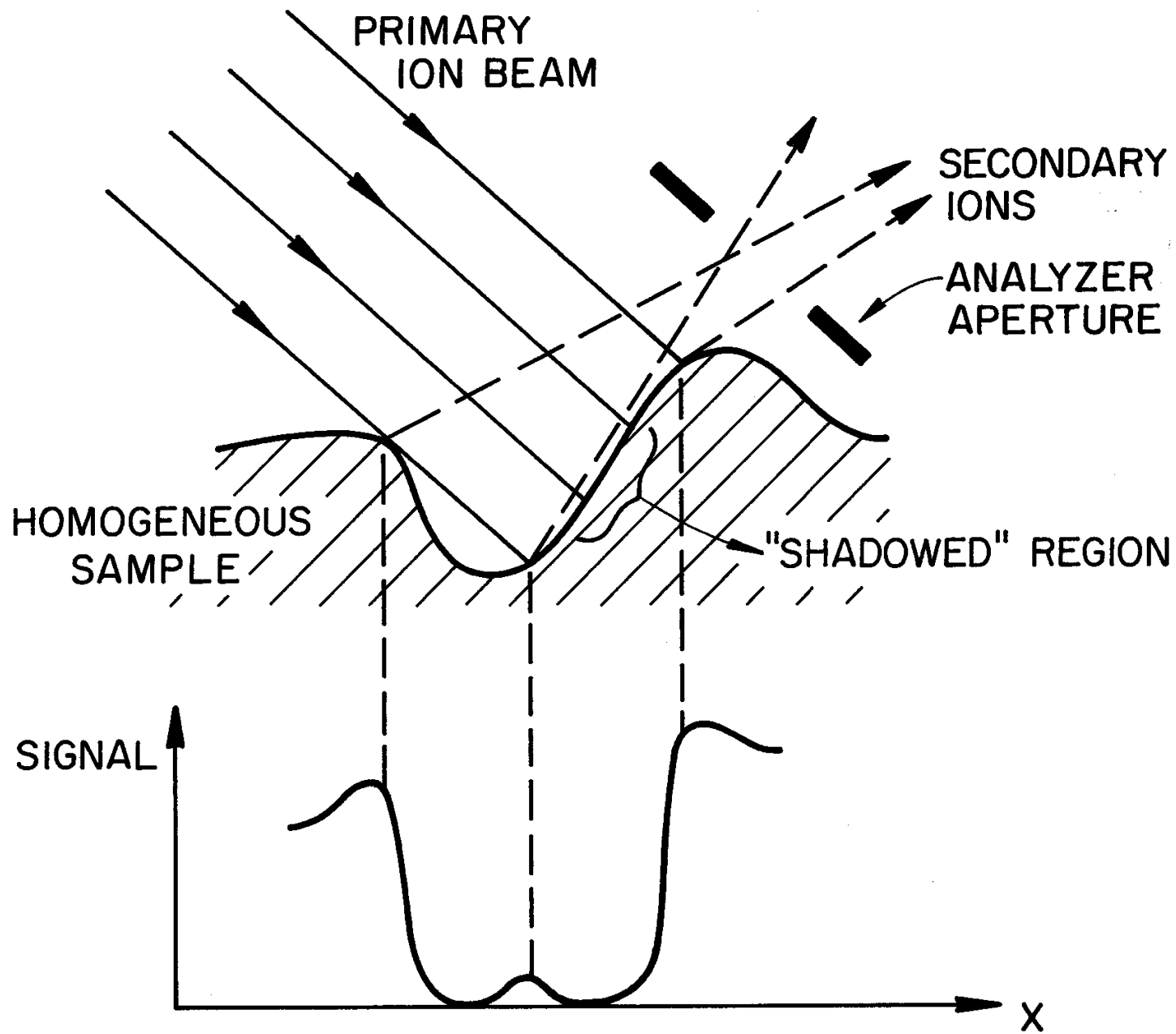
schematic

det can
energy detect
in time

cont



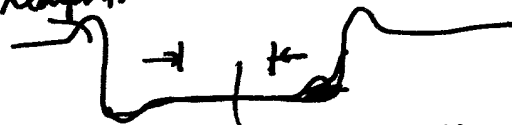




ASIDE 1 when depth profiling, tricks to eliminate artifacts due to redeposition, etc —

⚡ allows one to measure smaller concentrations

— analyze from smaller area than pulsed redeposit



sample this only by gated gating

see
doc cam
pic

~~~~~  
what is spatial resolution attainable?

1. depth  $\sim 5$  nm

2. lateral  $\sim 10-50$  nm

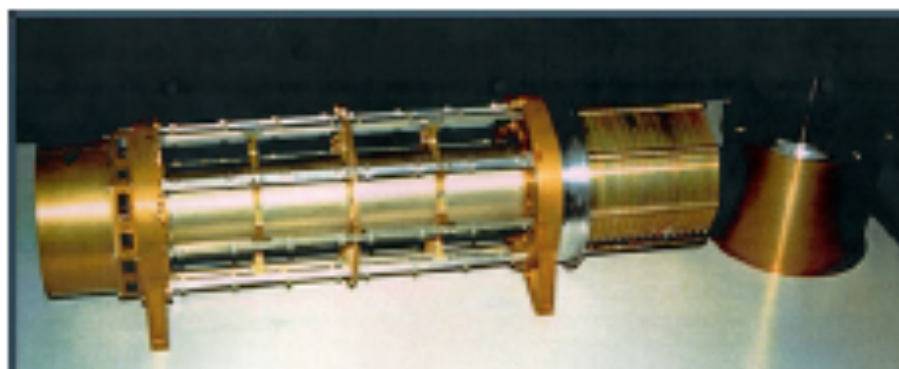
in addition:

3. all chem. elements can be observed, even H

4. separate isotopes can be measured  
(better than res. than RBS)

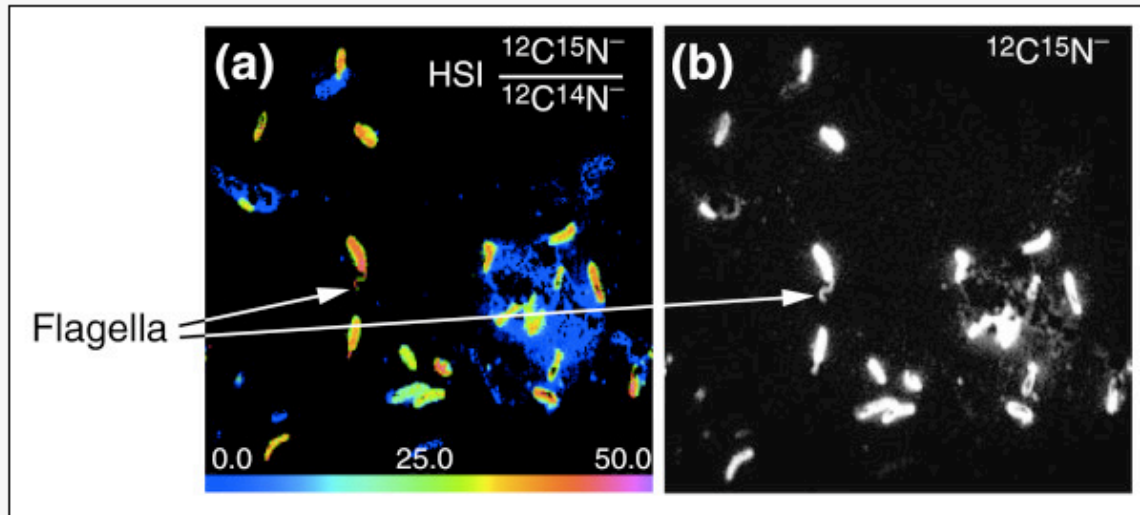
5. destructive

- ▶ The COSIMA instrument onboard the European Space Agency's Rosetta spacecraft will be the first instrument to determine the composition of cometary dust with SIMS in 2014.



<http://sci.esa.int/sci.es/en/www/obj/jsp/boilerplate006.jsp?ID=12124>

# Secondary Ion Microscopy in Biology



**Figure 2**

MIMS images of *Teredinibacter turnerae*, a nitrogen-fixing bacterium inhabiting a shipworm gill. The  $^{12}\text{C}^{15}\text{N}^-/^{12}\text{C}^{14}\text{N}^-$  image (left) allows the uptake of  $^{15}\text{N}$  by these organisms to be quantified. On the right, the image of  $^{12}\text{C}^{15}\text{N}^-$  ions shows that sensitivity is sufficient to image the flagellum of this bacterium (arrowed) even though the flagellum diameter is estimated from the ion signal to be only about 10 nm. From [1].

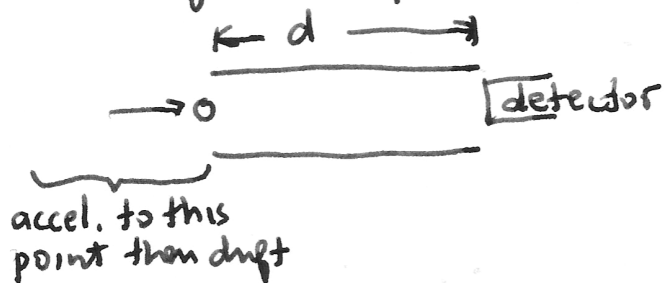
1. Lechene C, Hillion F, McMahon G, Benson D, Kleinfeld AM, Kampf JP, Distel D, Luyten Y, Bonventre J, Hentschel D, Park KM, Ito S, Schwartz M, Benichou G, Slodzian G. High-resolution quantitative imaging of mammalian and bacterial cells using stable isotope mass spectrometry. *J Biol.* 2006;5:20. doi: 10.1186/jbiol42. [[PMC free article](#)] [[PubMed](#)] [[Cross Ref](#)]

# Time of Flight Characterization (TOF)

potential energy  $E_p = QV$   
↳ charge

kinetic energy  $E_k = \frac{1}{2} m v^2$

send ions through a "drift" tube. field free space



$\therefore QV = \frac{1}{2} m v^2$  at tube entrance

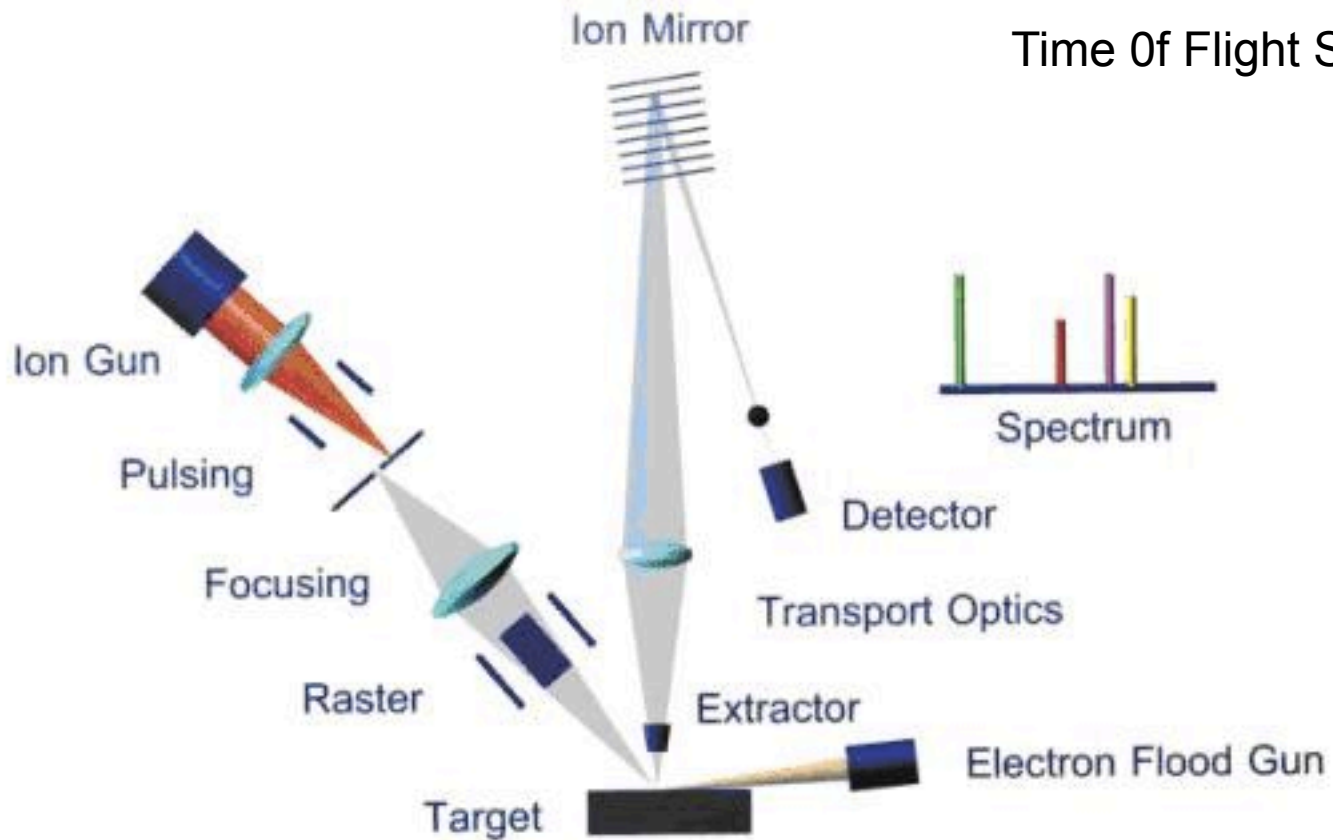
time through drift tube  $t = d/v$  ← ion velocity

$\therefore QV = \frac{1}{2} m \left( \frac{d}{t} \right)^2 \Rightarrow t = \frac{d}{\sqrt{QV}} \sqrt{\frac{m}{Q}}$

∥ traverse time related to mass/charge ∥



## Time Of Flight SIMS



Schematic diagram of the CAMECA IonTOF ToF-SIMS instrument.

# Time of flight SIMS

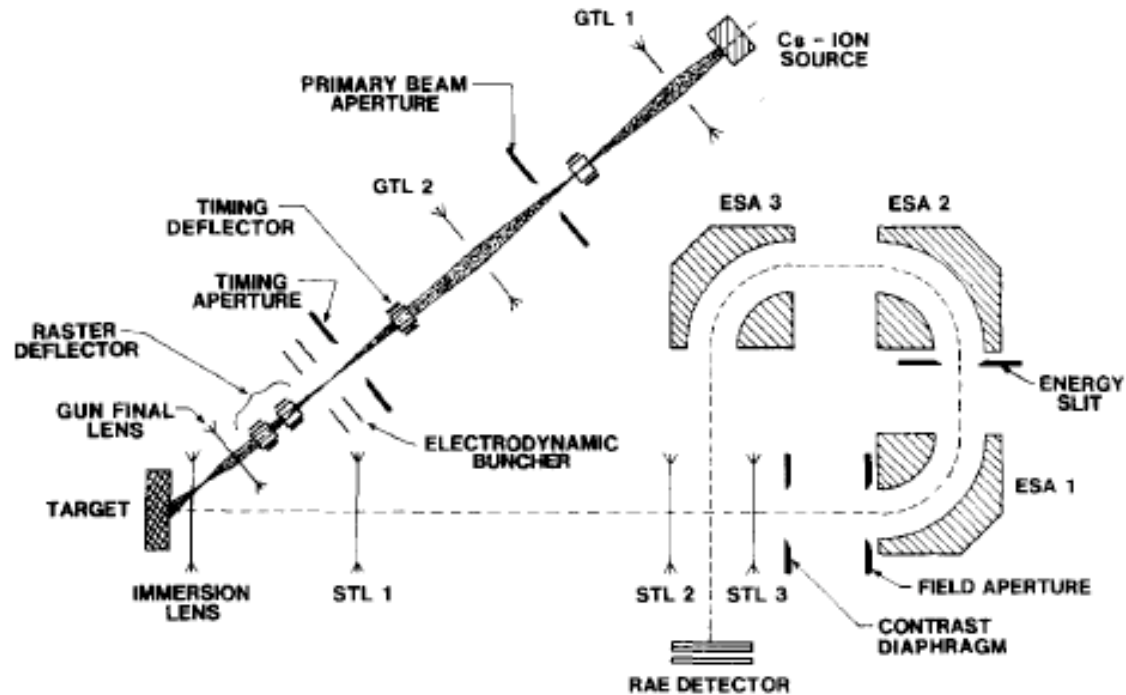
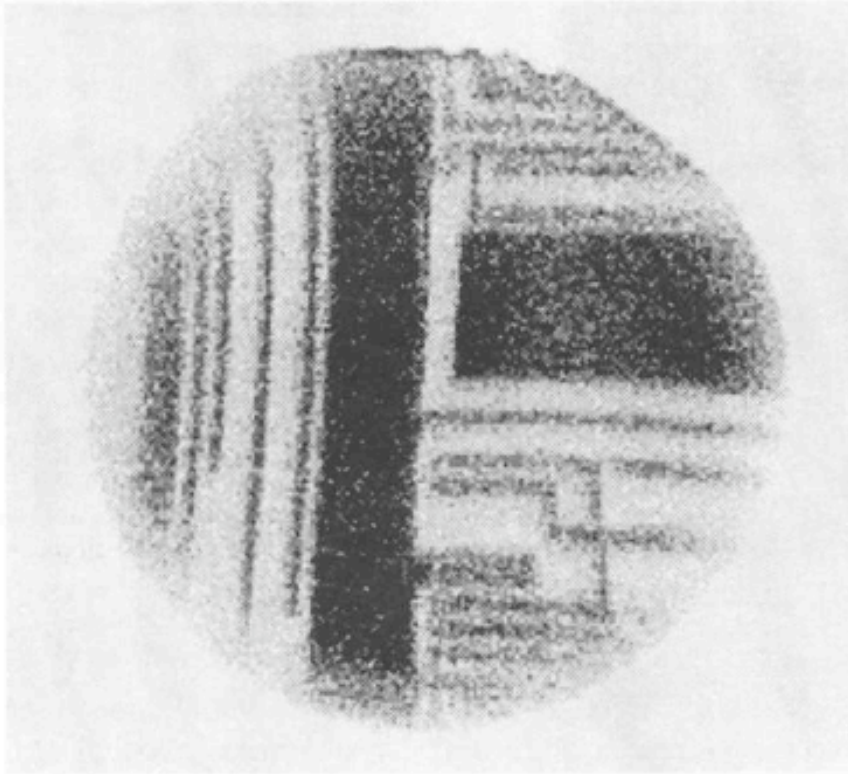


Figure 1. Schematic diagram of the Time-of-Flight secondary ion microscope.

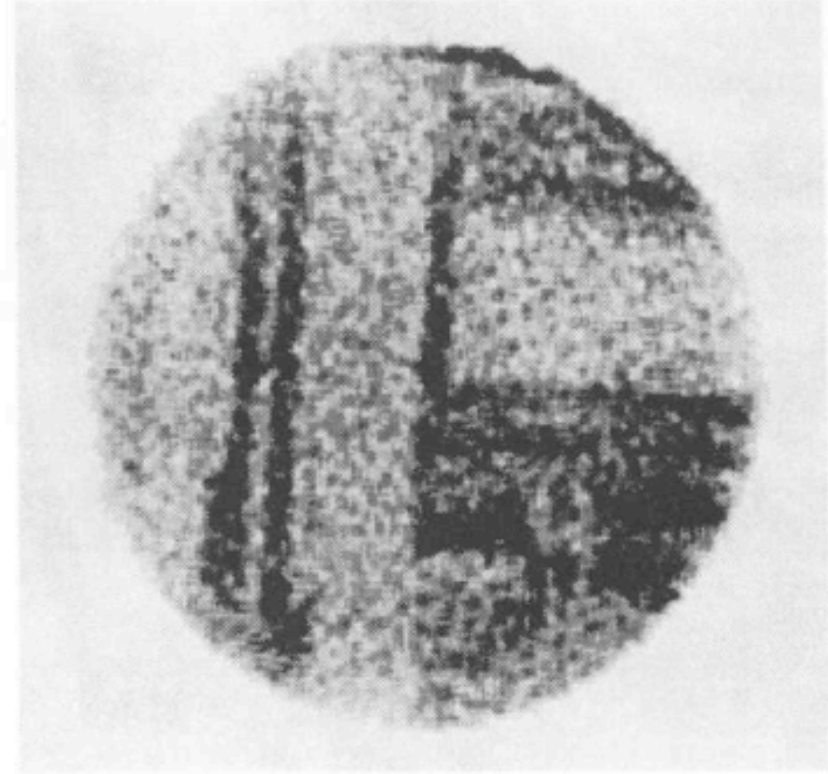
From: Schueler, et.al. Vacuum. 41(1-4).  
1990.p.1661-1664.

# Time of Flight SIMS Microscopy

a)  $\text{Al}^+$  ion image



b)  $\text{Si}^+$  ion image



**Figure 5.** Secondary ion images of  $\text{Al}^+$  (a) and  $^{28}\text{Si}^+$  (b) taken from a  $250\ \mu\text{m}$  diameter surface region of an Al-Si integrated circuit. The transmission reduction due to lateral energy filtering was approximately 30% for elemental secondary ion species.

From: Schueler, et.al. Vacuum. 41(1-4).1990.p.1661-1664.

# Commercial SIMS Instrument/ CAMECA



## Ion Microprobe, CAMECA NanoSIMS 50L

The Microanalysis Center for Geochemistry and Cosmochemistry at Caltech took the delivery of a CAMECA NanoSIMS 50L in last December and finished the installment and on-site tests in this past April. Now the instrument is fully operational and running well. The CAMECA NanoSIMS 50L is a new ion microprobe, developed for trace element and isotopic analysis of ultra-fine features. Among the unique new features offered by the NanoSIMS 50L are: 1) The ability to extend the SIMS analysis to extremely small areas or volumes ( $\sim 35$  nm size in cesium,  $\sim 150$  nm in oxygen) while maintaining extremely high sensitivity at high mass resolution (HMR). This derives from the revolutionary coaxial optical design of the ion gun and secondary ion extraction, and from a new design of the magnetic sector mass analyzer. 2) The capability of simultaneously measuring up to 7 masses (ions), ensuring more efficient and precise isotopic ratios from the same small volume, or better ion image superimposition in imaging mode. Along with the 7 electron multiplier (EM) detectors, 4 Faraday cups (FC) are also installed on the Caltech NanoSIMS 50L, enabling to achieve the precision and external reproducibility of isotope ratio measurements down to the low sub-permil level.





# Nano SIMS

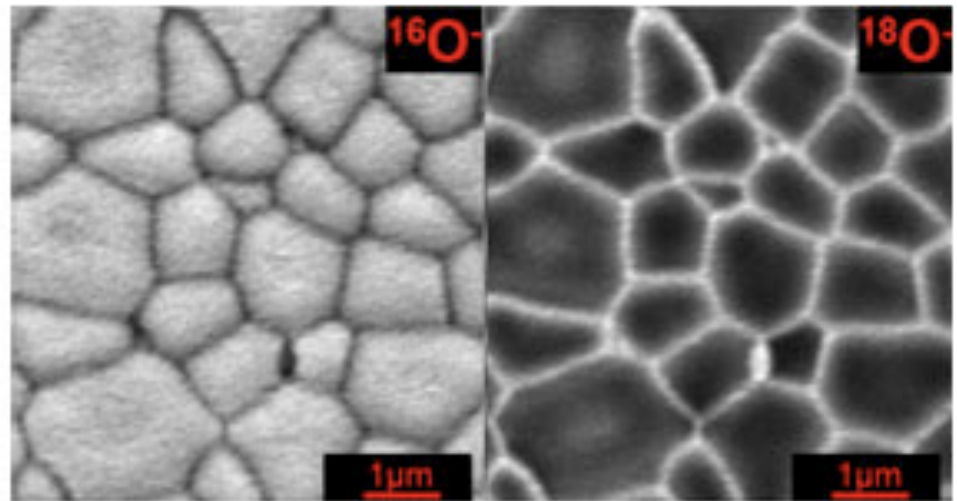
## Segregation and diffusion of elements in YAG (Yttrium Aluminium Garnet)

The **NanoSIMS** is used here to contribute to the understanding of the segregation and diffusion of elements in polycrystalline materials.

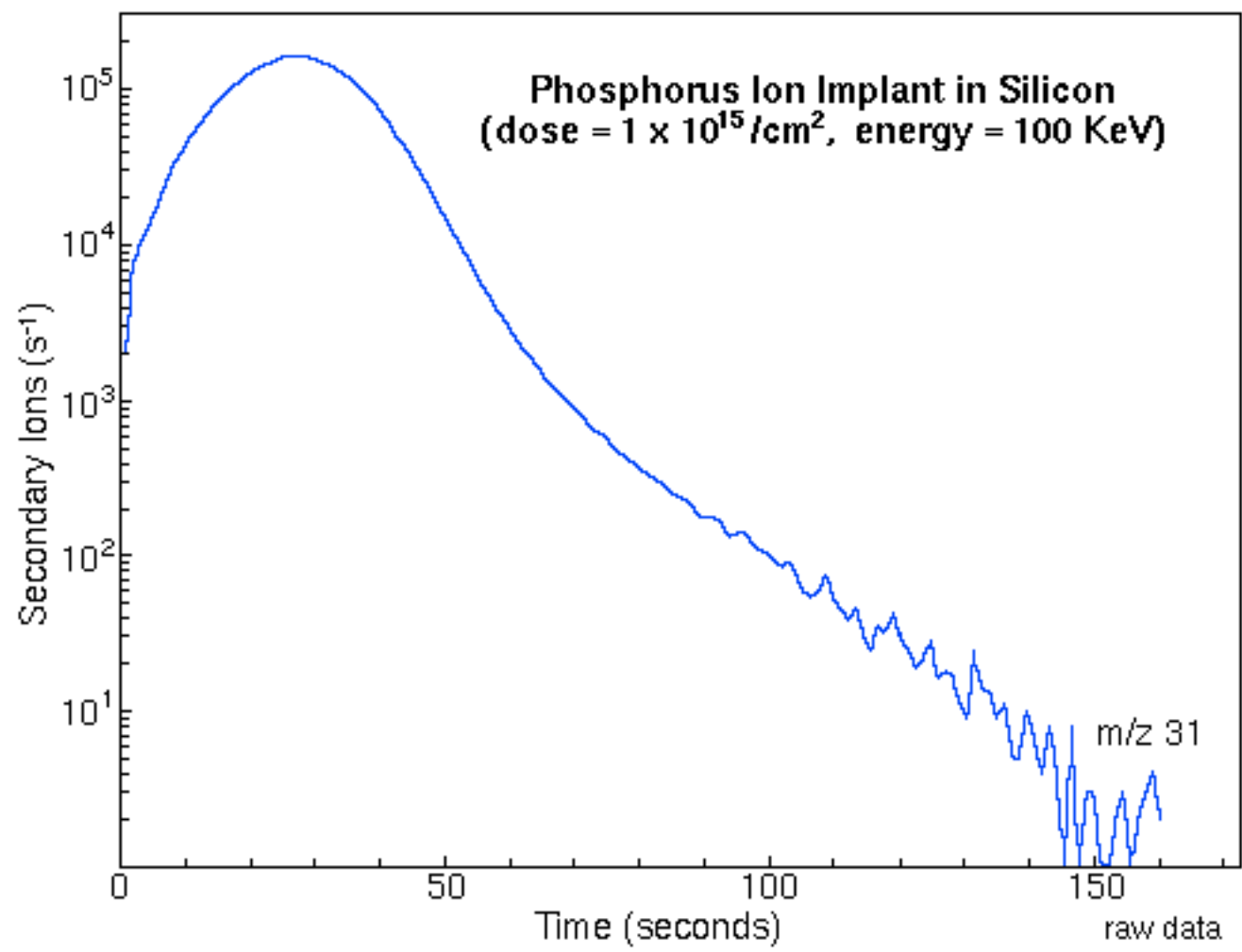
In the first example, the strategy is to make **use of  $^{18}\text{O}$  stable isotope tracers** in order to image and quantify the incorporation of oxygen.

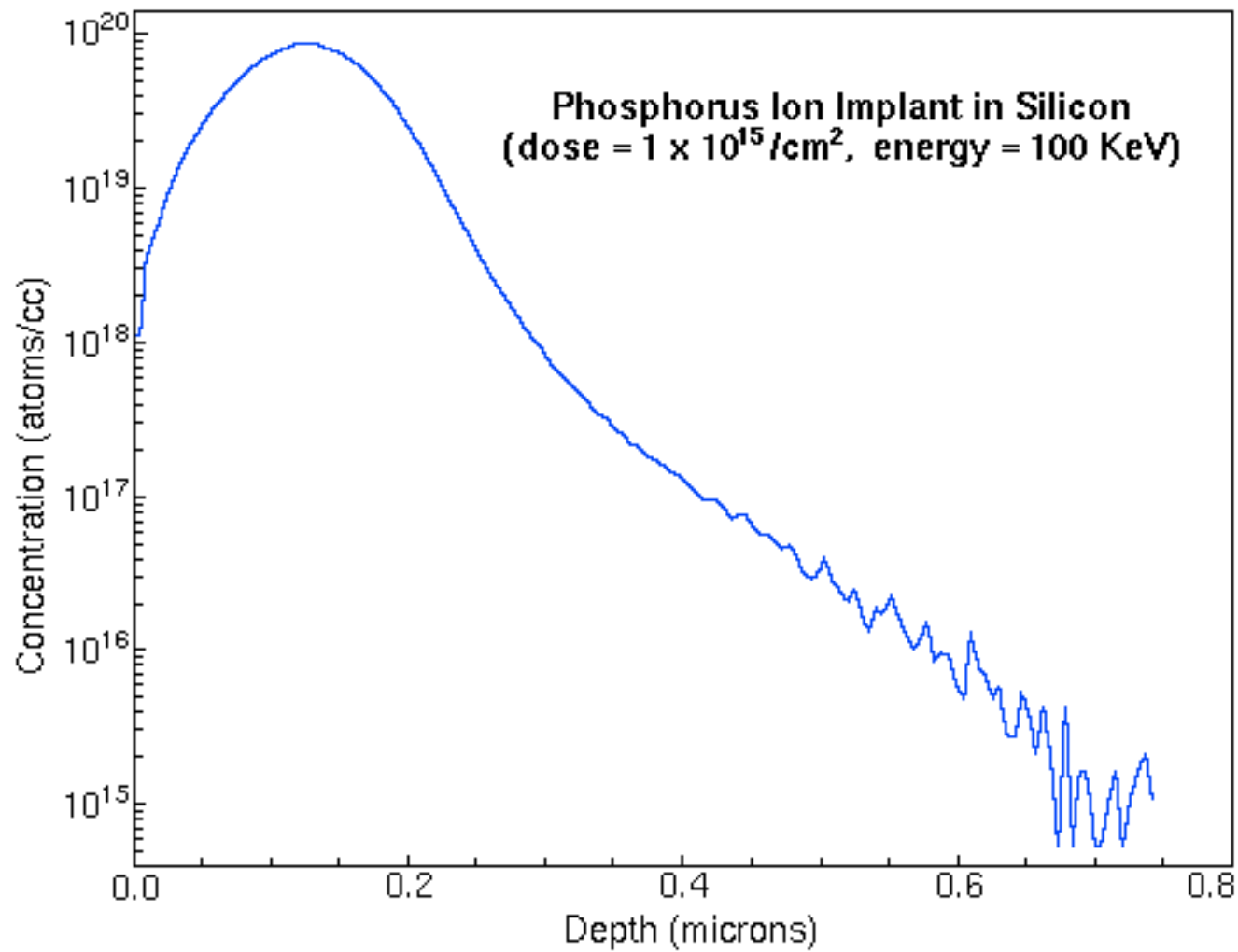
A sintered transparent YAG (yttrium aluminum garnet) is oxidized in a  $^{18}\text{O}_2$  atmosphere at  $1400^\circ\text{C}$ .

The  $^{18}\text{O}$  oxygen atoms diffuse inside the ceramics through the grain boundaries as evidenced from the two complementary images of  $^{16}\text{O}$  (base element of the oxide) and  $^{18}\text{O}$  (incorporated during the oxidation sequence).



Dr. Hajime Haneda, NIMS, Tsukuba, Japan



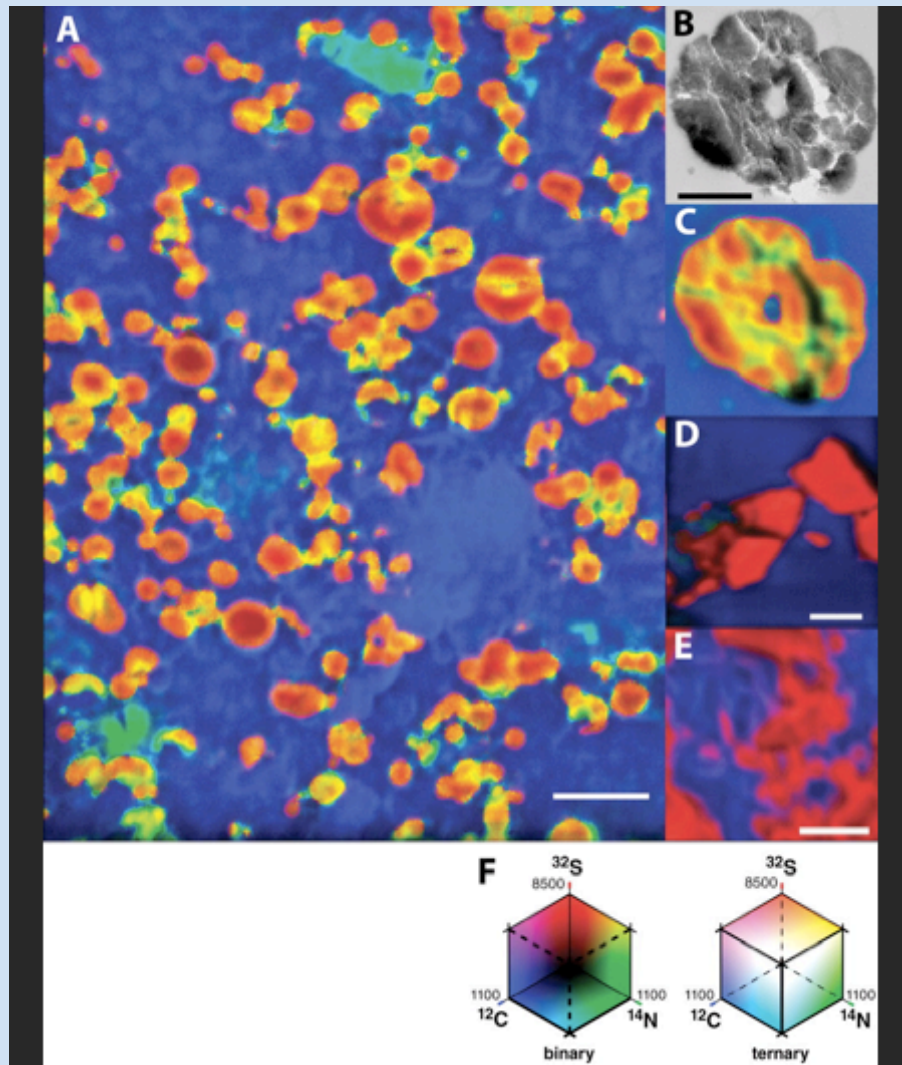




REPORT

## Extracellular Proteins Limit the Dispersal of Biogenic Nanoparticles

John W. Moreau<sup>1,2,†</sup>, Peter K. Weber<sup>2</sup>, Michael C. Martin<sup>3</sup>, Benjamin Gilbert<sup>4</sup>, Ian D. Hutcheon<sup>2</sup> and  
Jillian F. Banfield<sup>1,4,5</sup>



## Some References for SIMS Microscopy

G. Slodzian. Surface Science.48.(1961)

R. Castaing and G. Slodzian. (1962) J. Microscopy.1.395-410  
(in French)

H. Liebl.(1967) J.A.P. 38.5277-5283.

R. Levi-Setti et. Al. (1986). Applied Surface Science.  
26.249-264, (H<sup>+</sup> probe)

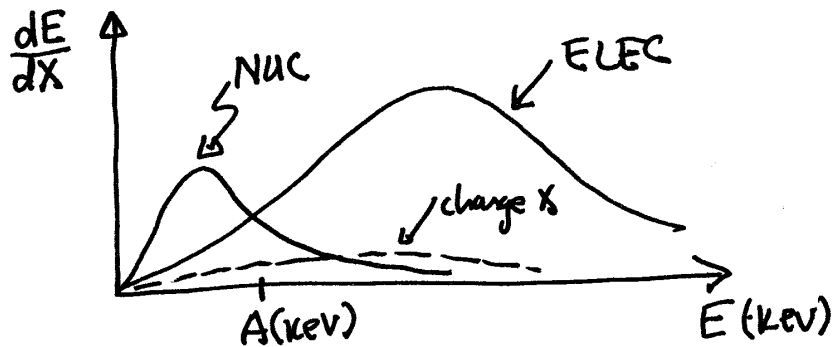
## ION INTERACTIONS

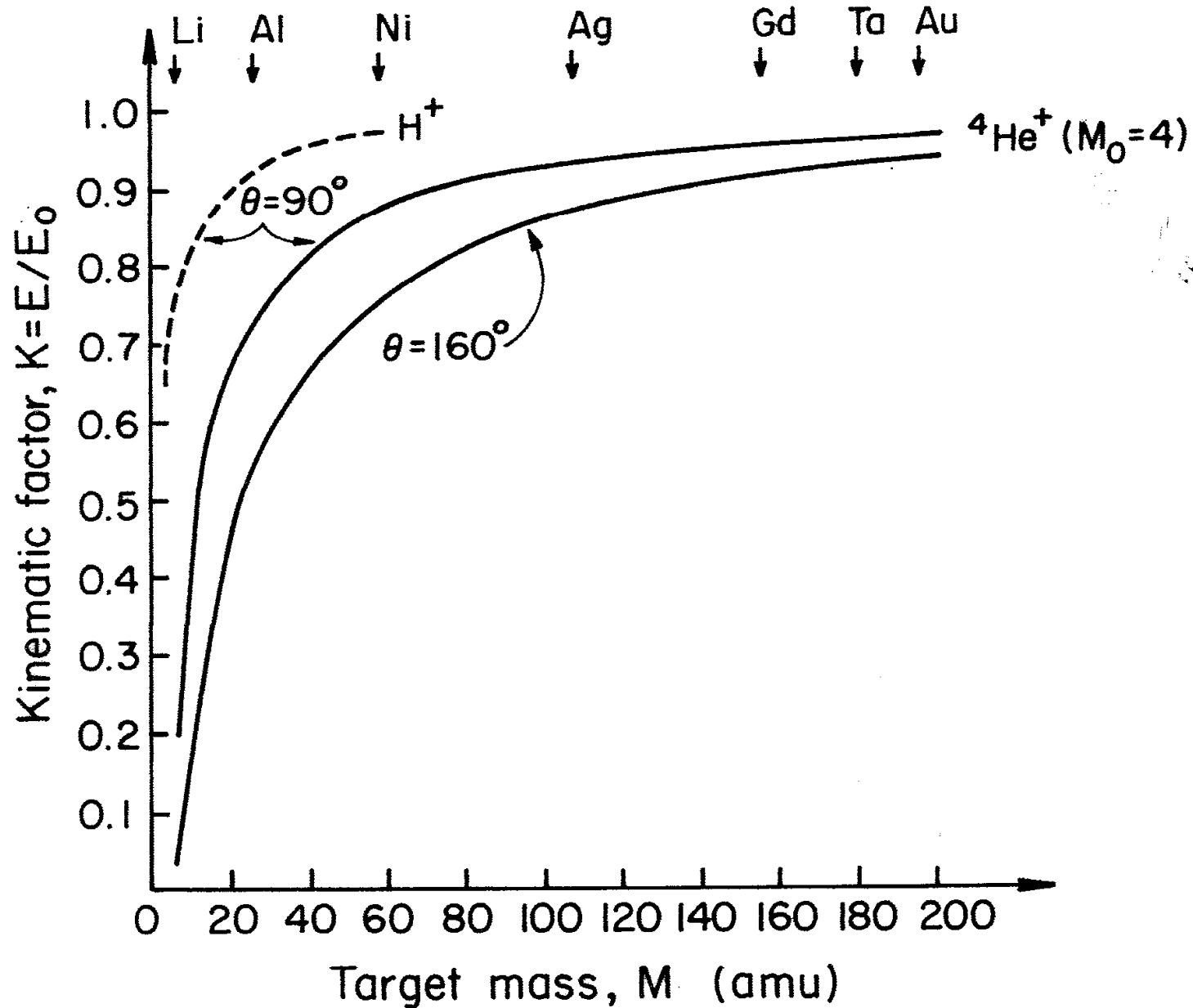
three main components

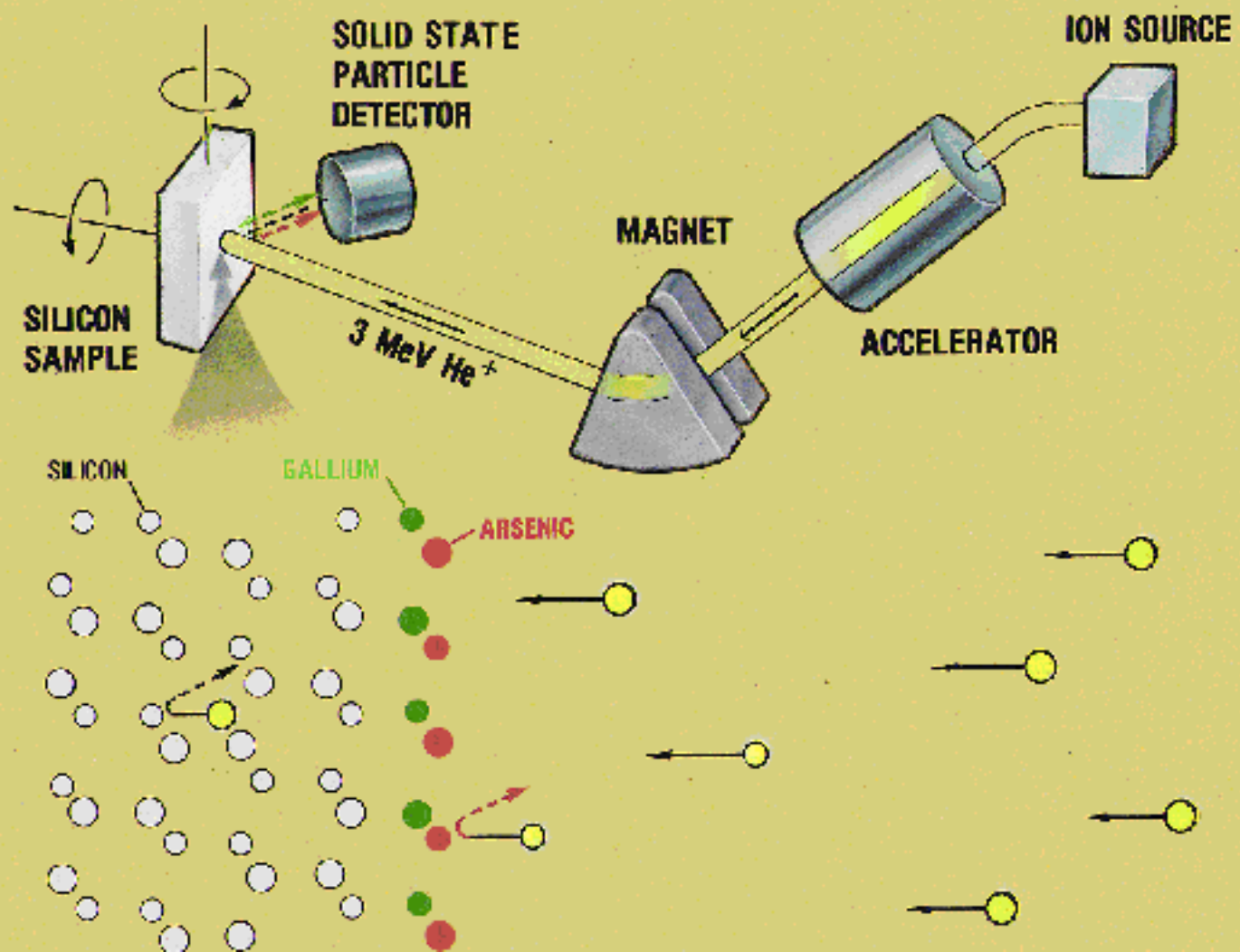
1. "nuclear" energy loss, dominates at low energies  
when  $E < A$  keV  
 $\uparrow$   
 at. wt. ions

2. electronic energy loss  
interactions with atomic electrons  
(like "inelastic" electron scattering)

3. charge exchange  
 $\lesssim 10\%$  of ~~the~~ total

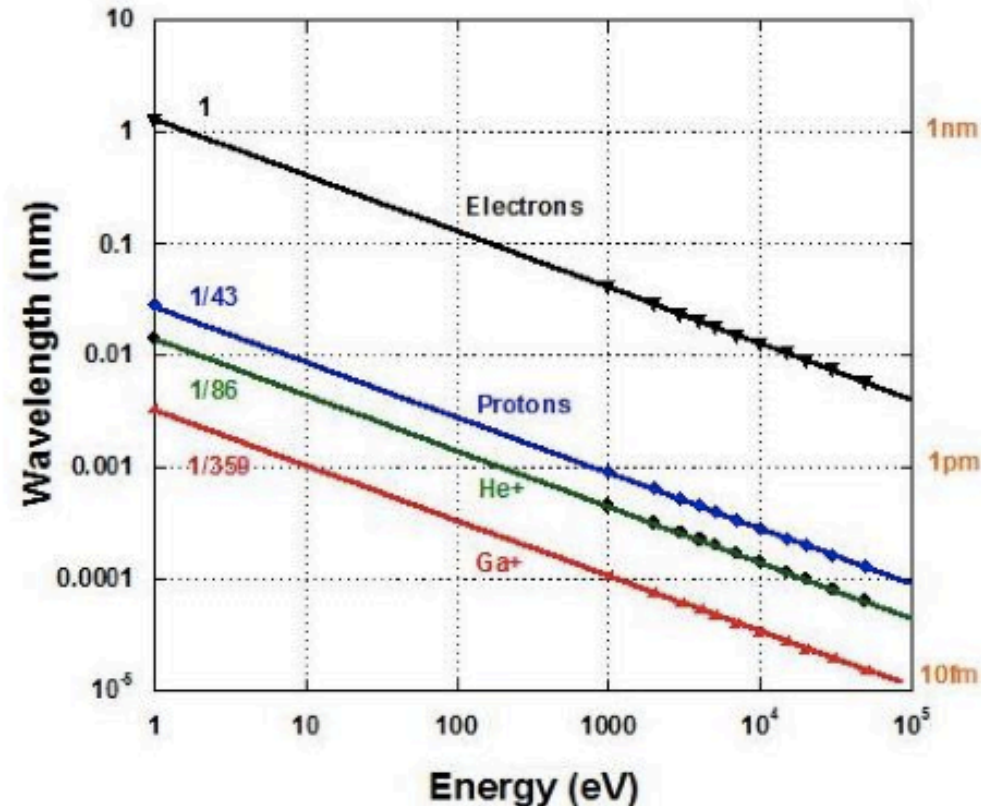






# Why use ions?

- Ions have wavelengths  $\lambda$  ~1% or less of that for electrons of same energy
- **Diffraction limited spot size ( $=\lambda/\alpha$ ) is negligible**
- Which ion to chose?
- $H^+$  (proton) medium iSE yield, low sputter damage. Used by R. Levi-Setti with success (1970s)
- $He^+$  - higher yields but more sputter. Good source available...



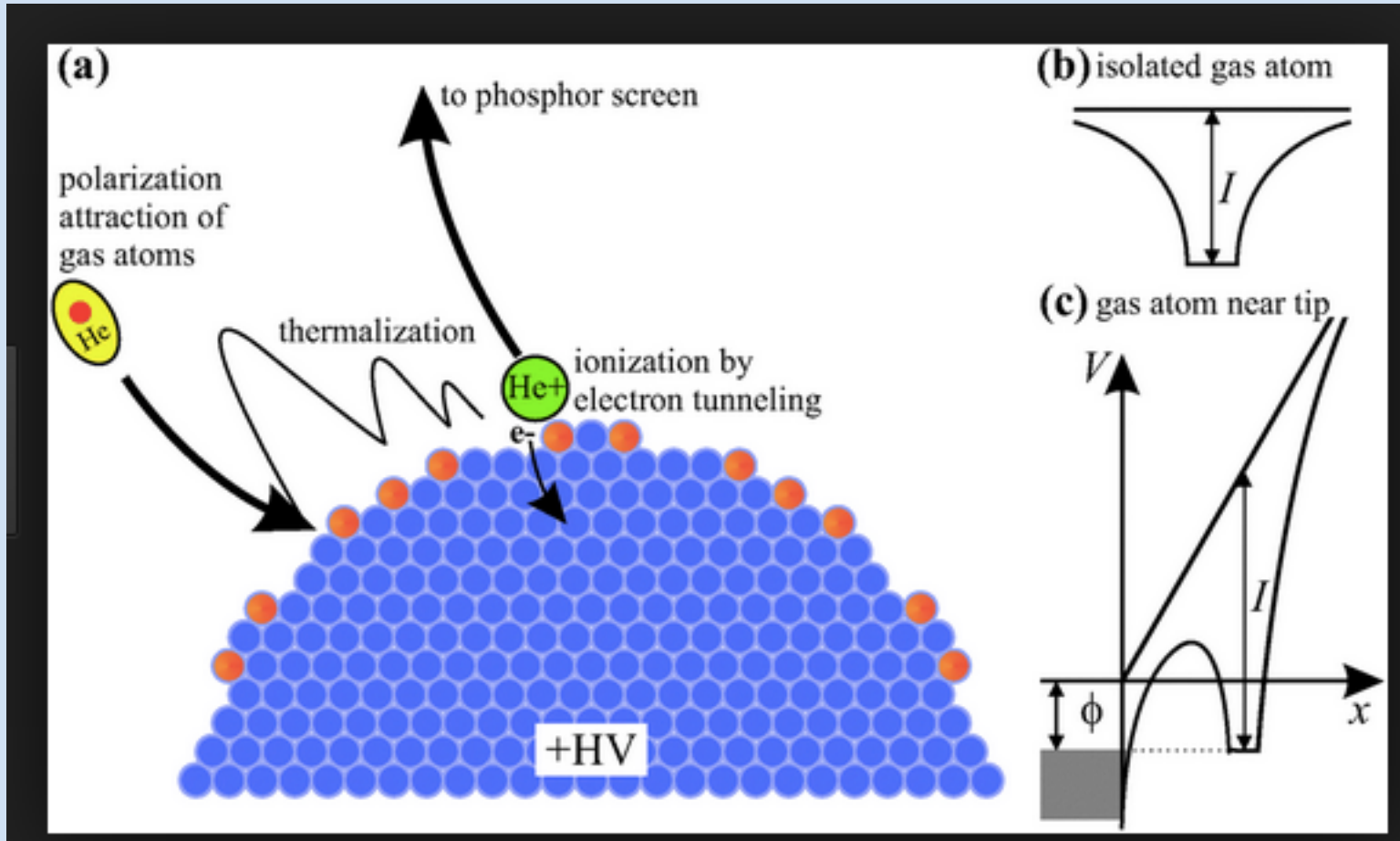
Wavelength ( $\lambda$  nm) of electrons, and protons, helium and gallium ions, as a function of energy

# He Ion Beam Microscope (combine SEM with RBS)

- The scanned beam is 35 keV  $He^+$  ions rather than electrons
- The scanned He ions are launched from a *single metal atom* on the ion source needle
- The helium ion source brightness of  $3.6 \times 10^9$  A/cm<sup>2</sup>str and a 1:1 column demagnification provide for:
  - ♦ 0.3 nm resolution at a 8 mm working distance
  - ♦ A 2.5  $\mu$ m depth of focus
- Two detectors collect  $He^+$  induced secondary electrons (*SE mode*) or Rutherford Backscattered Ions (*RBI mode*) to form two very different images simultaneously
- Ion induced SE's have an energy mean of  $\sim 2$ eV with  $\sim 1$ nm mean free path in conductors,
  - ♦ hence SE mode images are extremely *surface sensitive*
- Electron flood interlace, while HIM imaging, neutralizes any positive surface charge, hence:
  - ♦ *Samples need not be coated and*
  - ♦ Dielectrics can be imaged at the highest beam energy and resolution



# Field Ionization





# Field Ion Microscopy

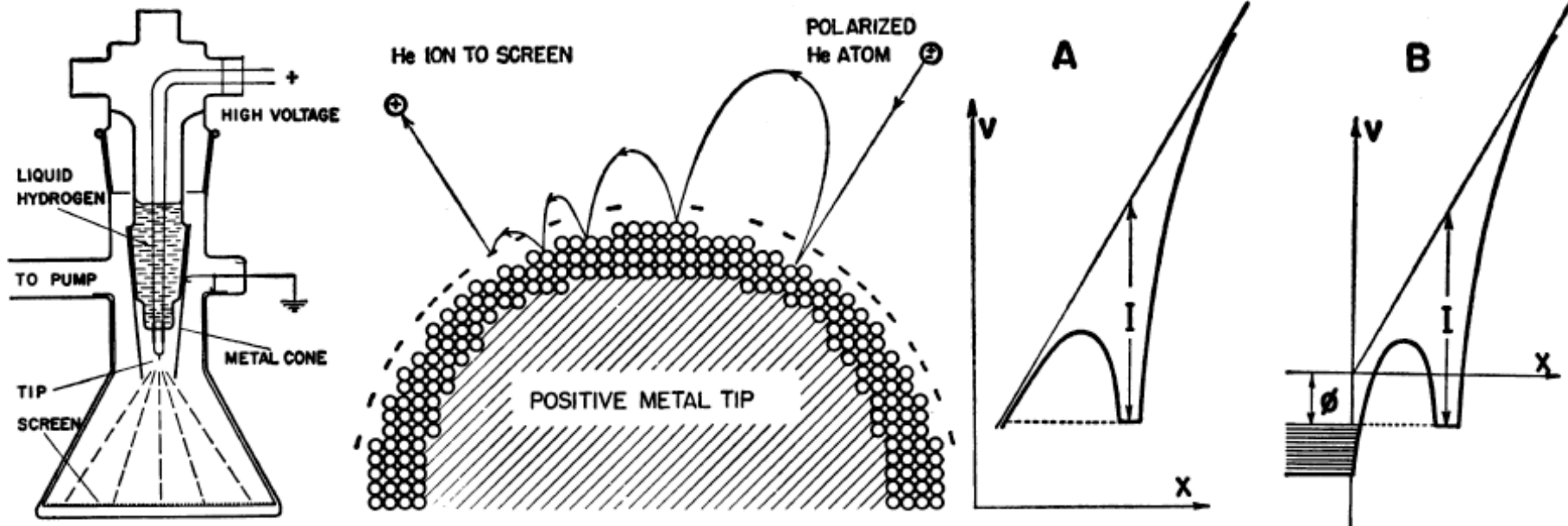


Fig. 1 (left). Schematic drawing of a field ion microscope. Fig. 2 (center). A polarized helium atom is attracted to the metal tip, is slowed down in a number of hops, and is ionized in the ionization zone (0.2 Å thick) over a protruding atom. The helium ion is accelerated toward the screen (see Fig. 1). Fig. 3 (right). (A) The potential funnel of an atom in a strong electric field; (B) the potential of an electron near the metal surface. Field ionization of the gas atom occurs when the electron from the ground state tunnels along the dotted line into the metal to the left.

E. W. Mueller, SCIENCE.149(#3684).1960.p.591-601..

# Field Ion Micrograph: Platinum Tip

E. W. Mueller, *Science*. 149,591 (1965)

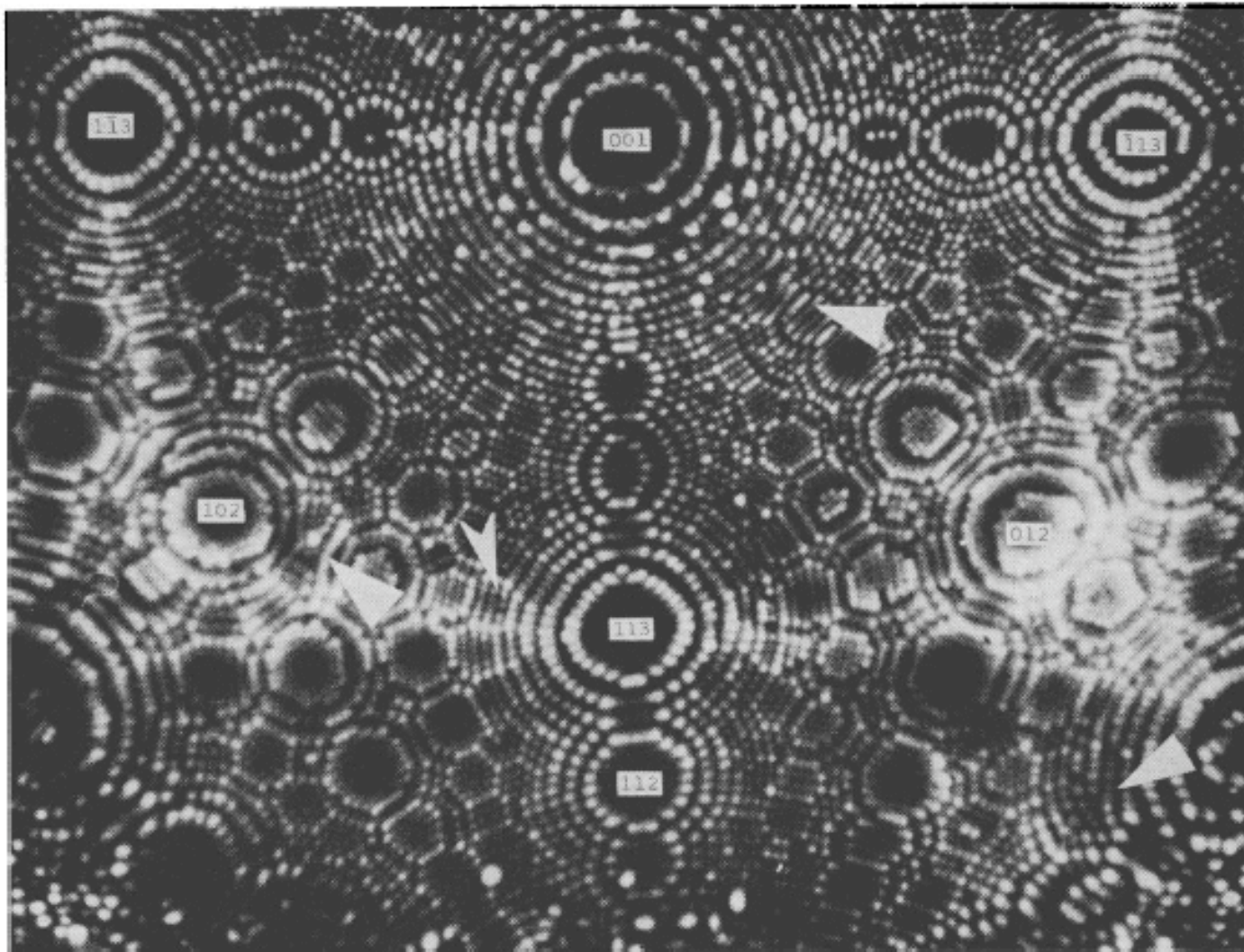


Fig. 13. Platinum crystal with a number of dislocations. The dislocation core near the (102) plane is decorated with an impurity atom, appearing as a bright spot. A vacancy is seen near the (113) plane.

AMERICAN VACUUM SOCIETY  
CLASSICS

# FIELD EMISSION AND FIELD IONIZATION

Robert Gomer

## Field Emission and Field Ionization

Book by Robert Gomer

Market: Students and researchers in vacuum and surface science, microscopy, and semiconductor physics. This definitive work was based on four lectures presented at Harvard University in 1958. ... [Google Books](#)

**Originally published:** 1961

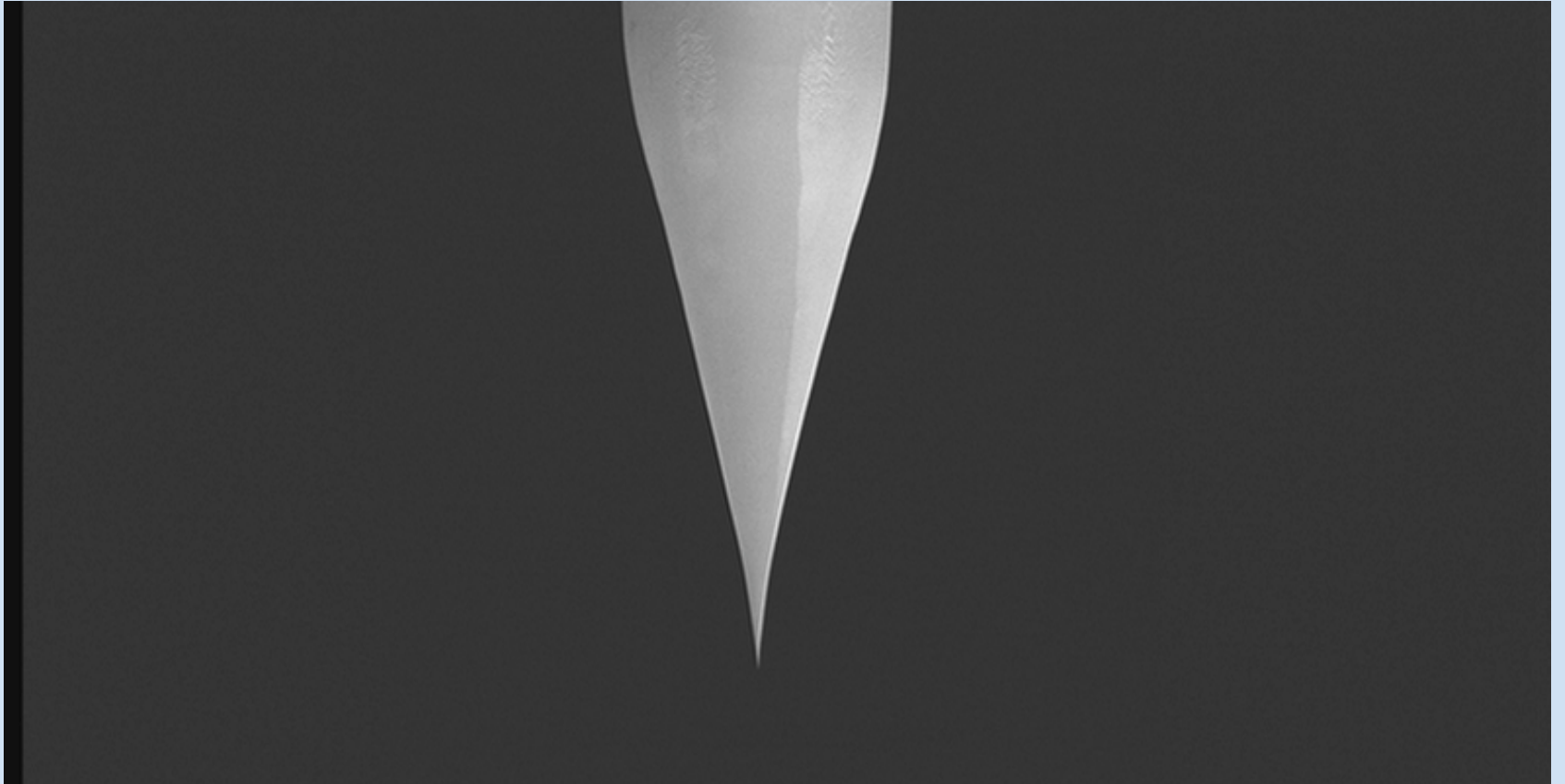
**Author:** [Robert Gomer](#)

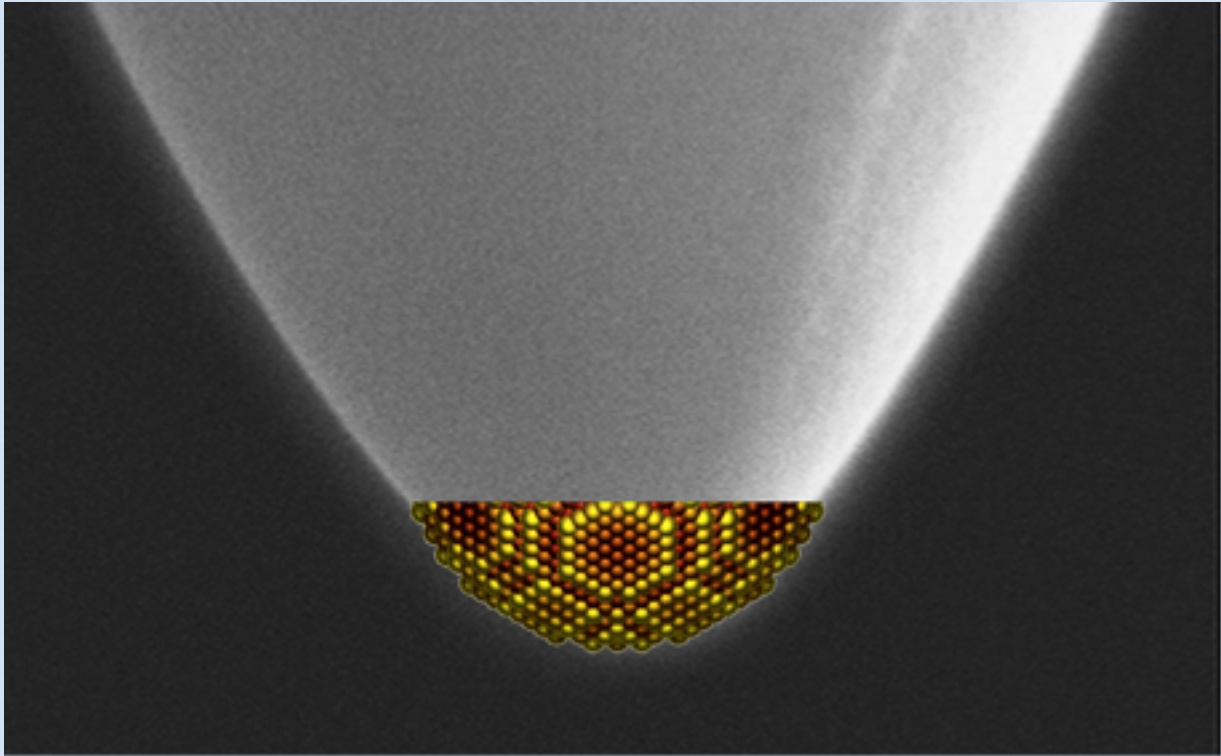
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CLASSICS

FIELD EMISSION  
AND FIELD  
IONIZATION

Robert Gomer

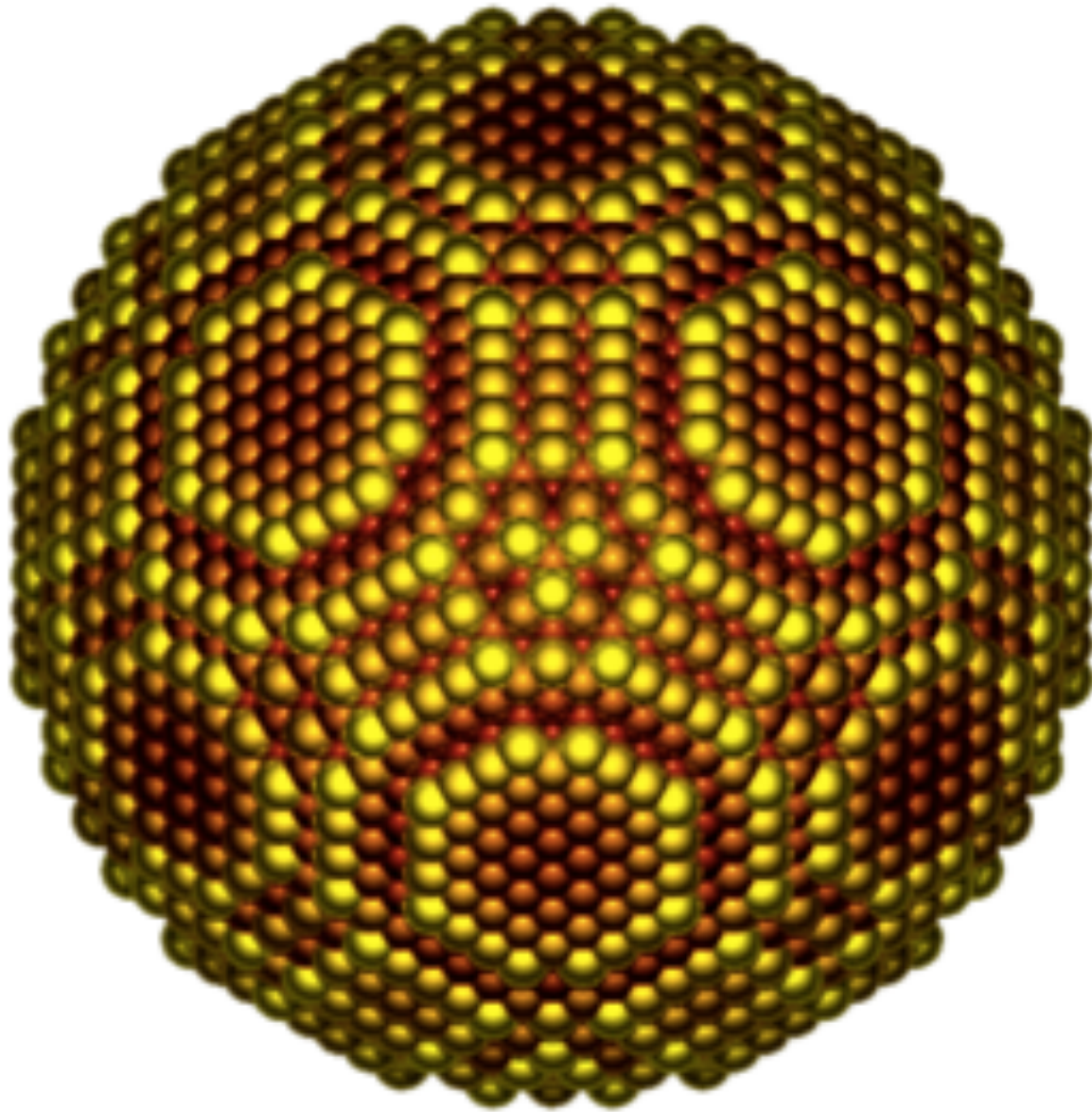
# Helium Ion Source



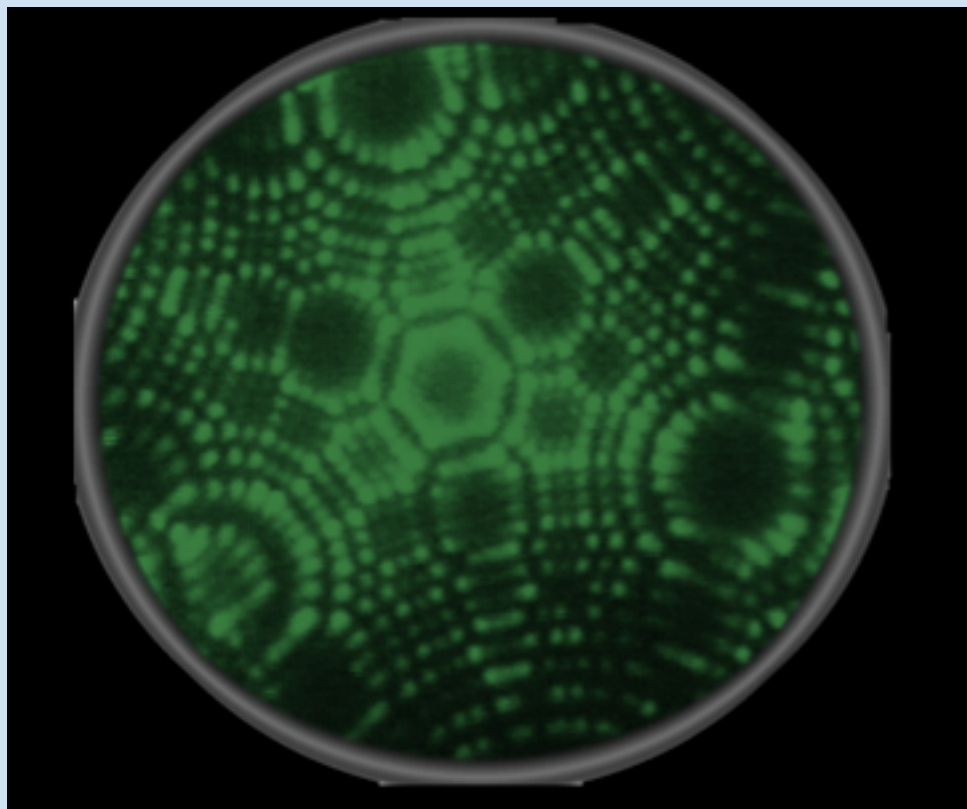




(111) W TIP (atomic view)



## Field Ion Micrograph of the Unshaped Emitter



Probe Size:

$$d_p = \sqrt{(M \cdot d_g)^2 + d_s^2 + d_c^2 + d_d^2}$$

Demagnified source:

$$d_{so} = M \cdot d_g$$

Spherical aberration:

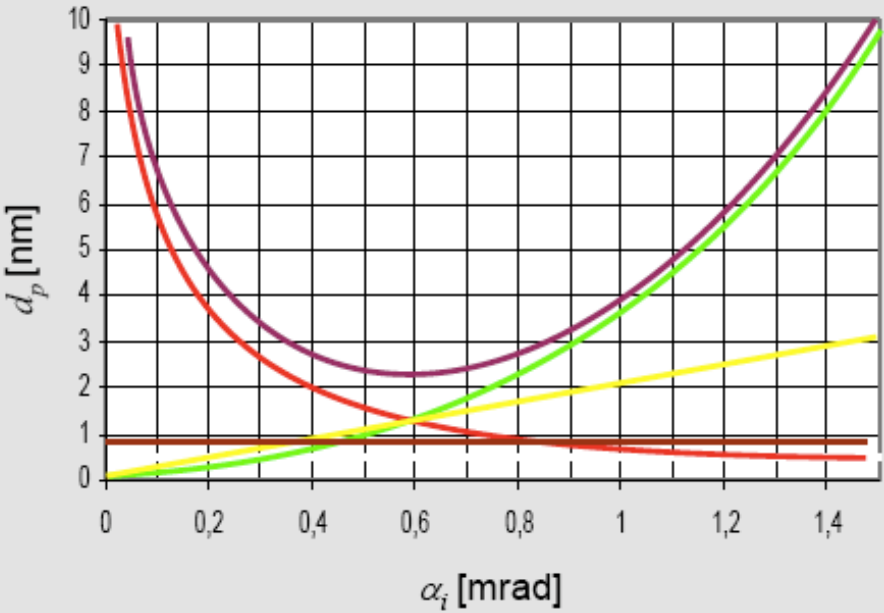
$$d_s = 0.5 C_s \alpha_i^3$$

Chromatic aberration:

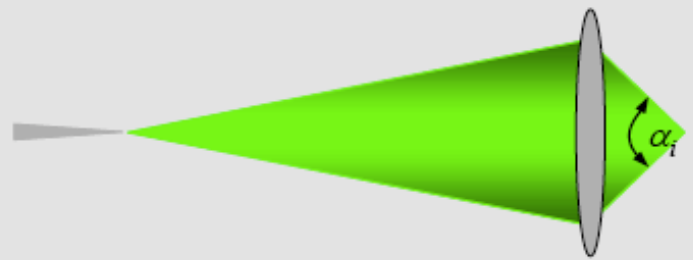
$$d_c = C_c \frac{\Delta U}{U} \alpha_i$$

Diffraction Error:

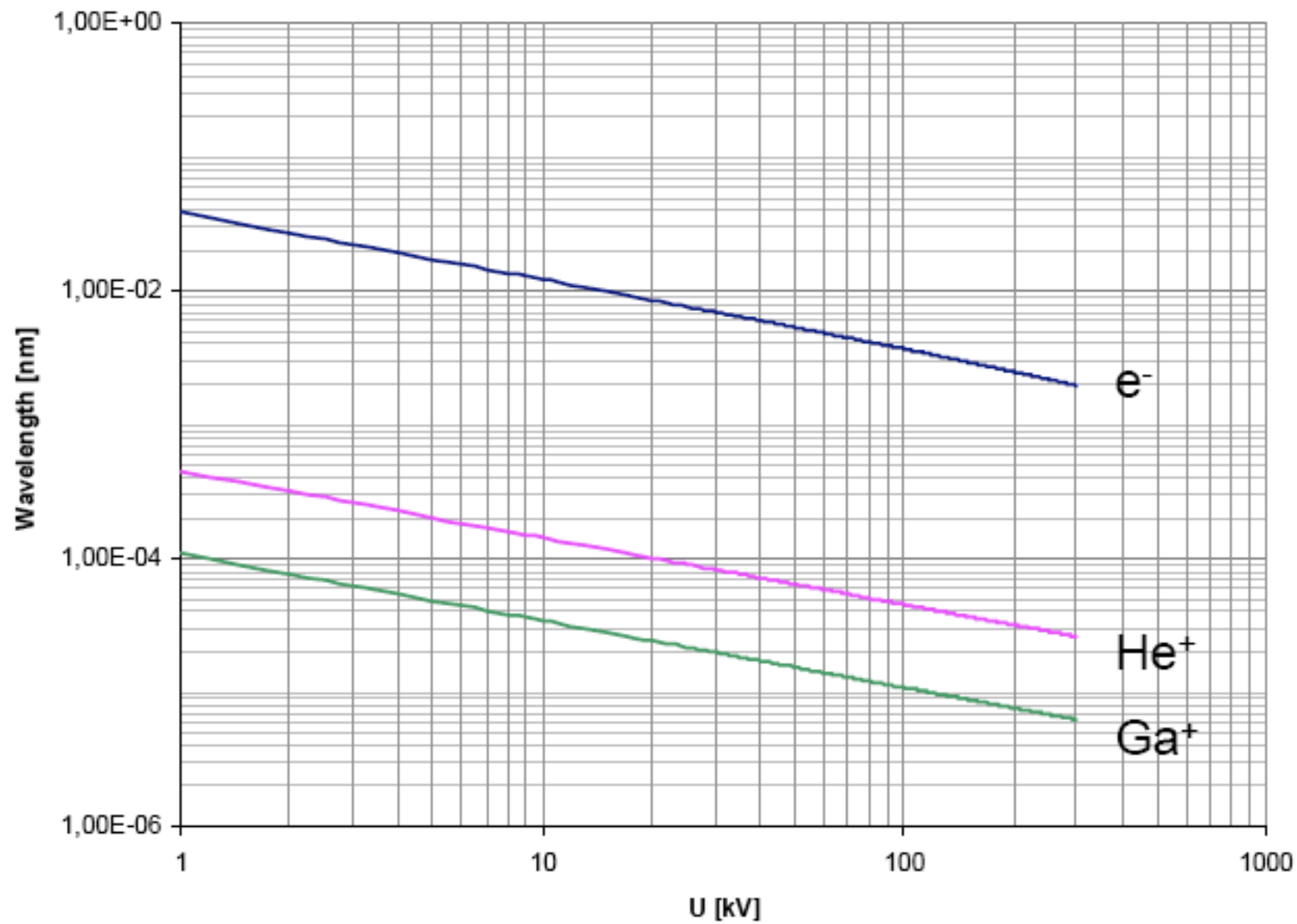
$$d_d = 0.6 \frac{\lambda}{\alpha_i}$$



Superposition of the aberration discs







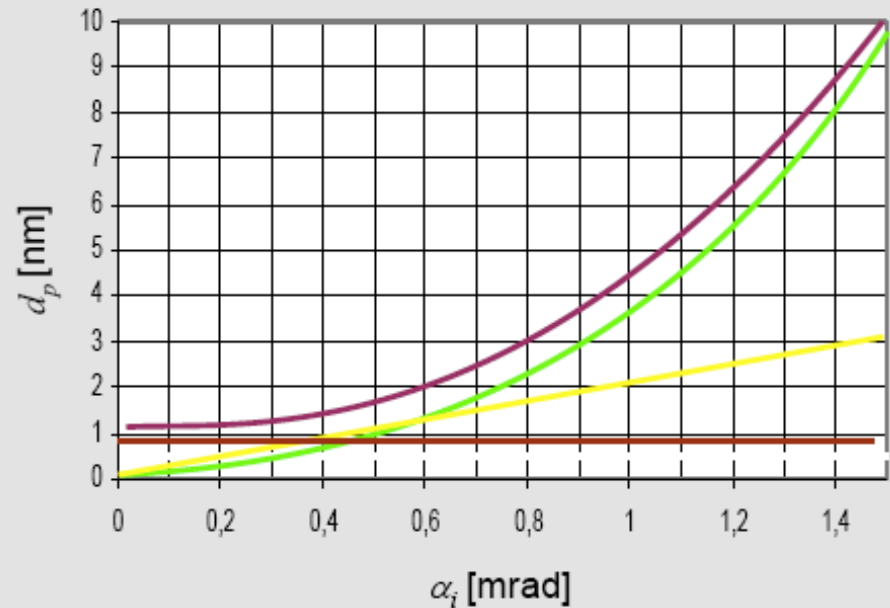
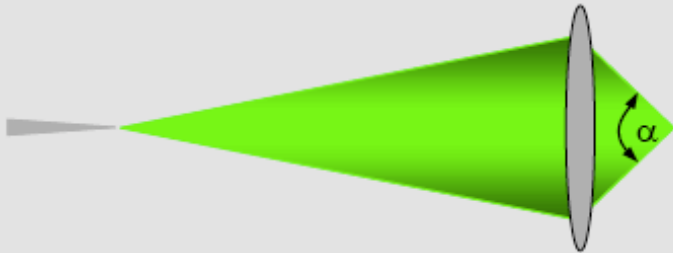
Probe Size:  $d_p = \sqrt{(M \cdot d_g)^2 + d_s^2 + d_c^2 + d_d^2}$

Demagnified source:  $d_{so} = M \cdot d_g$

Spherical aberration:  $d_s = 0.5 C_s \alpha_i^3$

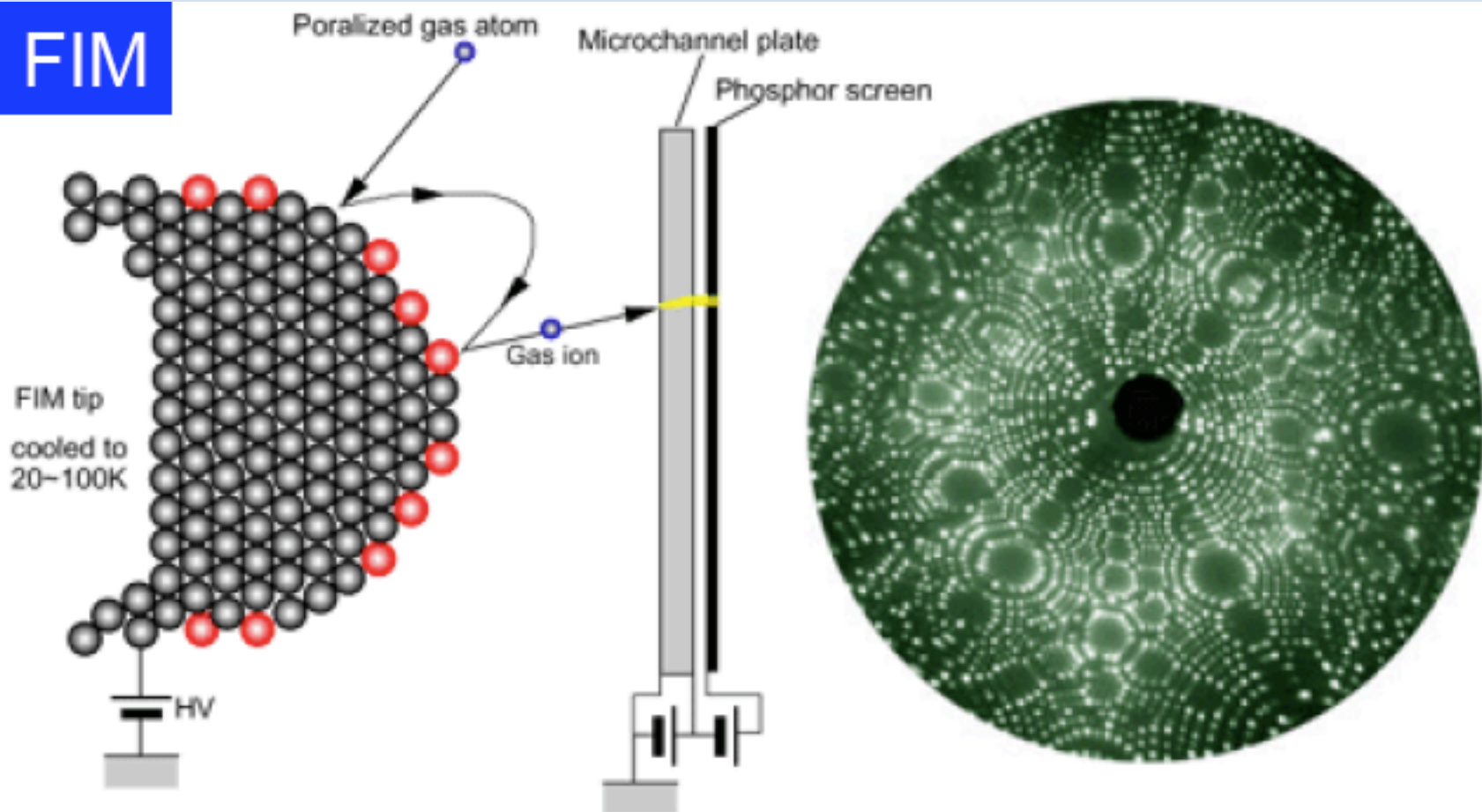
Chromatic aberration:  $d_c = C_c \frac{\Delta U}{U} \alpha_i$

~~Diffraction Error:  $d_d = 0.6 \frac{\lambda}{\alpha_i}$~~

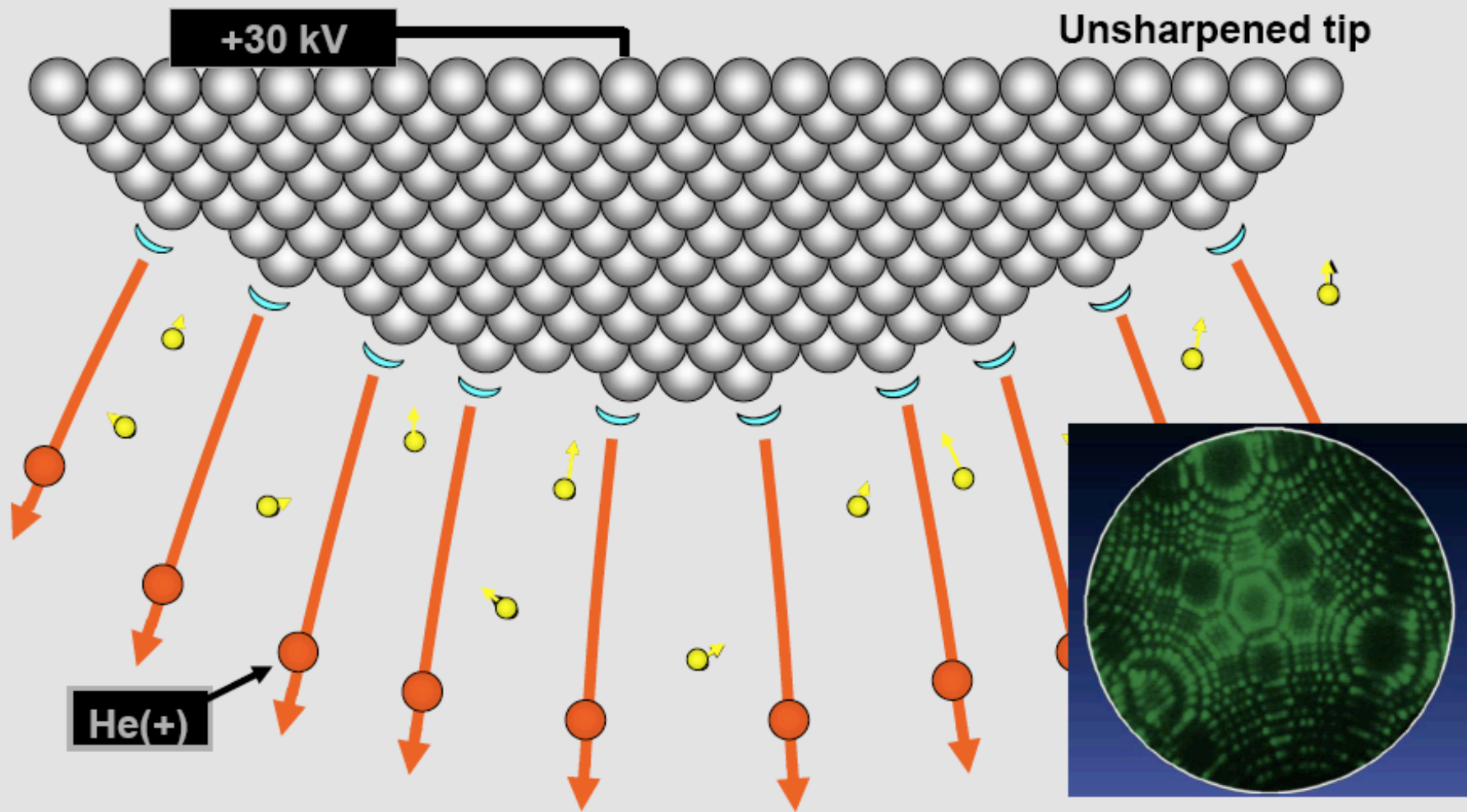


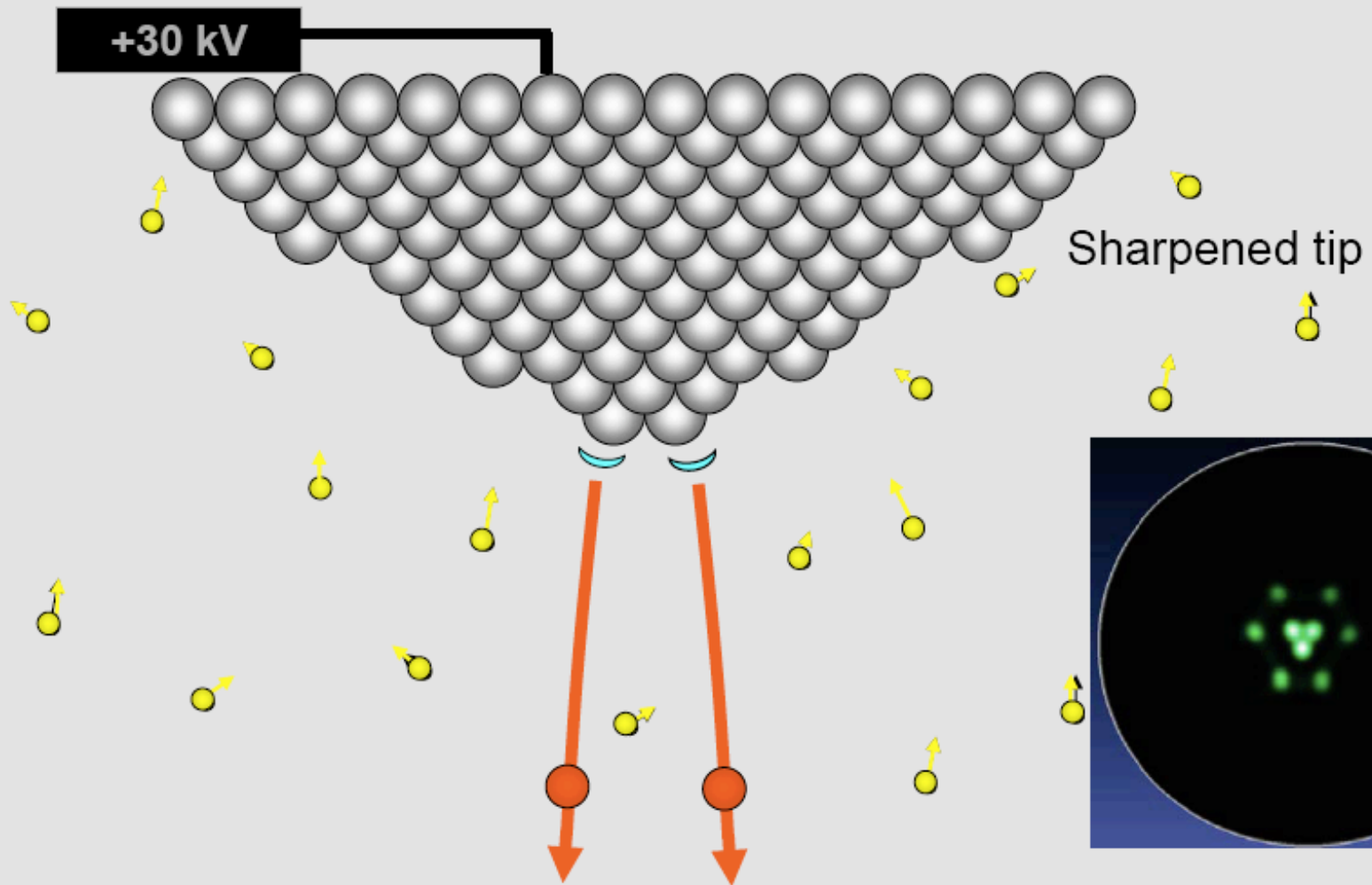
Superposition of the aberration discs

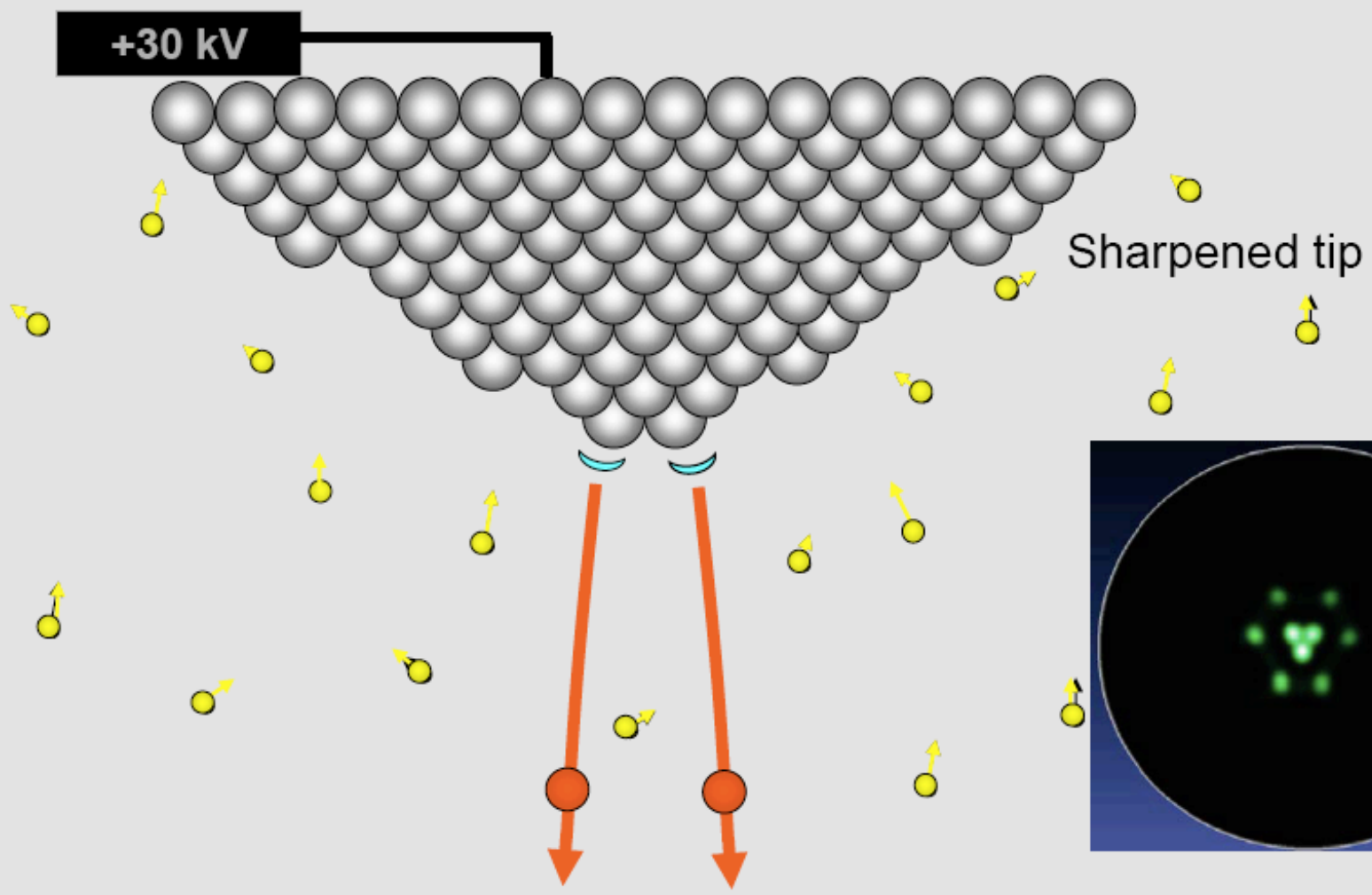
# FIM



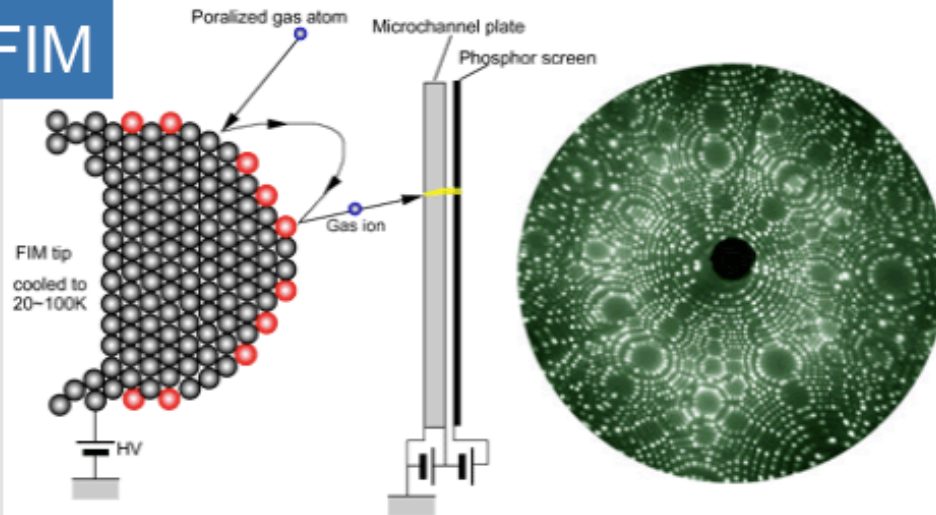
Schematics of a field ion microscope



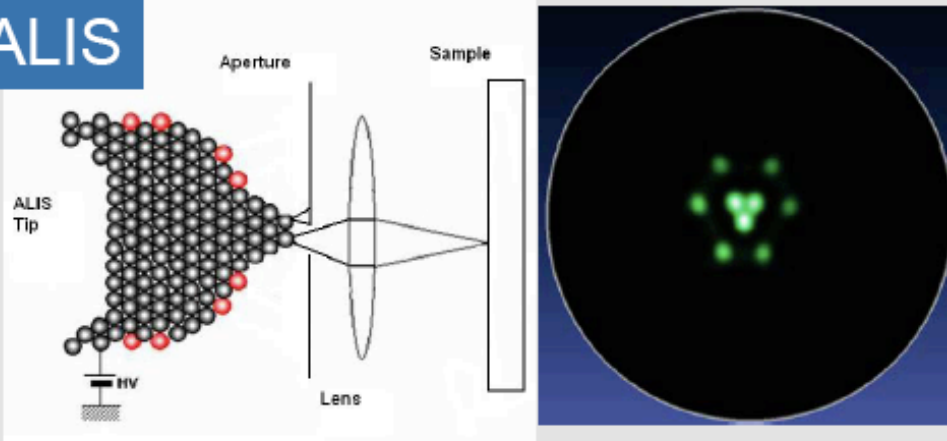




## FIM

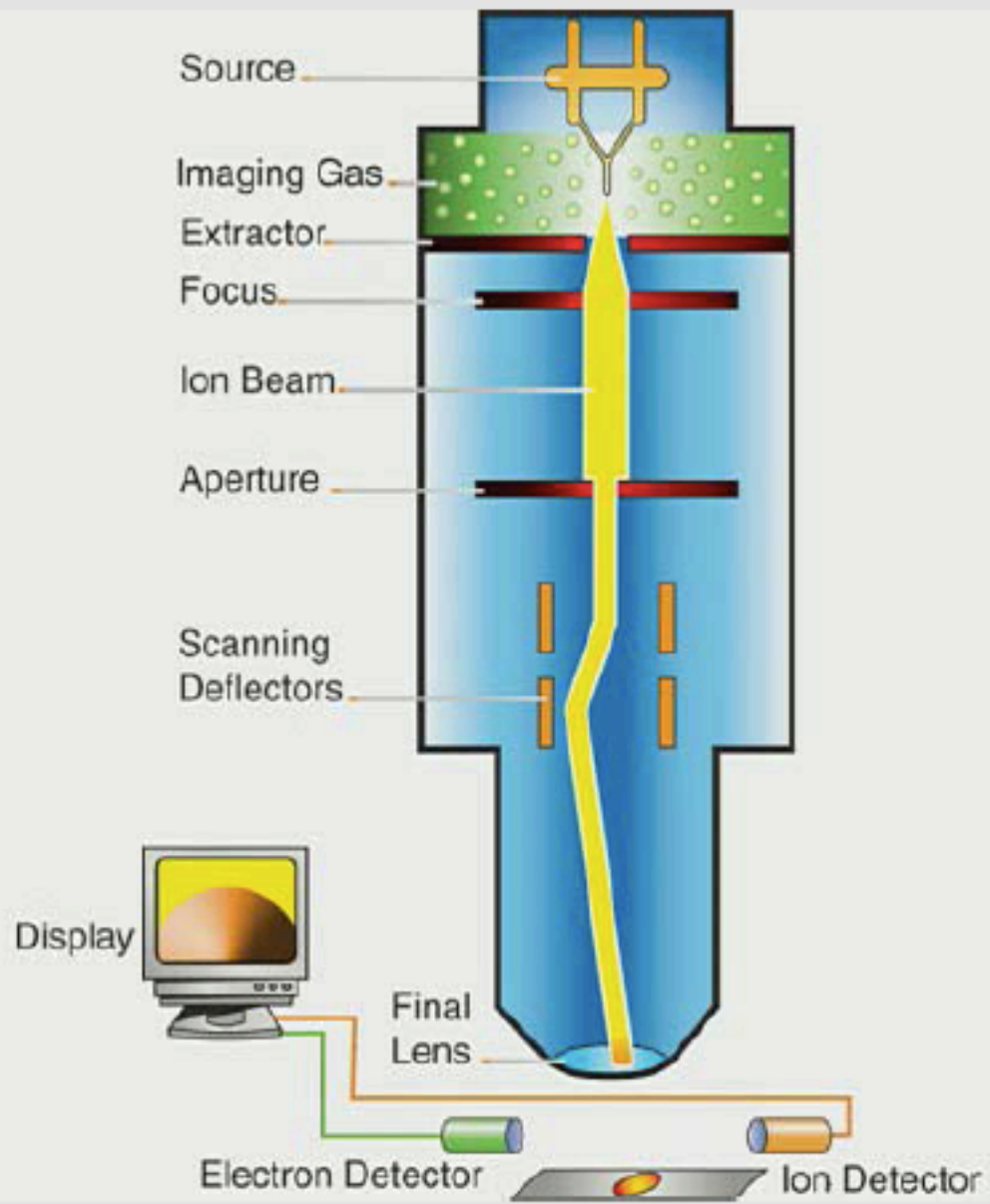


## ALIS



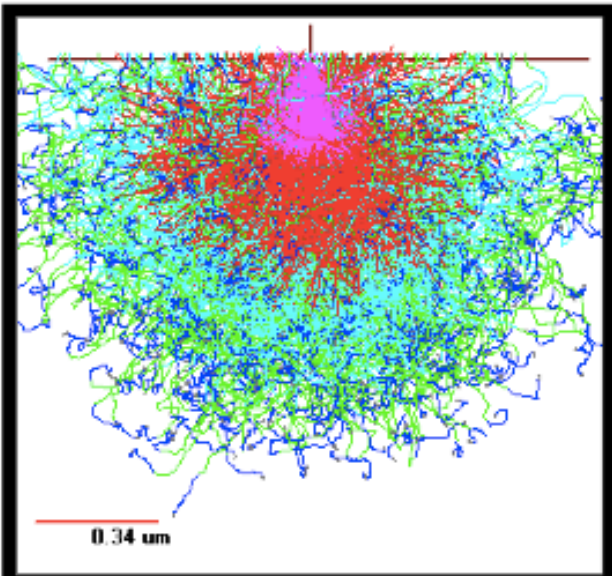
- FIM tip created via chemical etching
- ALIS tip formed with additional reshaping
- 3 atom shelf called the “trimer” created through field evaporation
- Single atom selected to form the final probe
- Results in a sub-angstrom virtual source with high brightness ( $4 \times 10^9$  A/(cm<sup>2</sup> sr)) and low energy spread (<1eV)



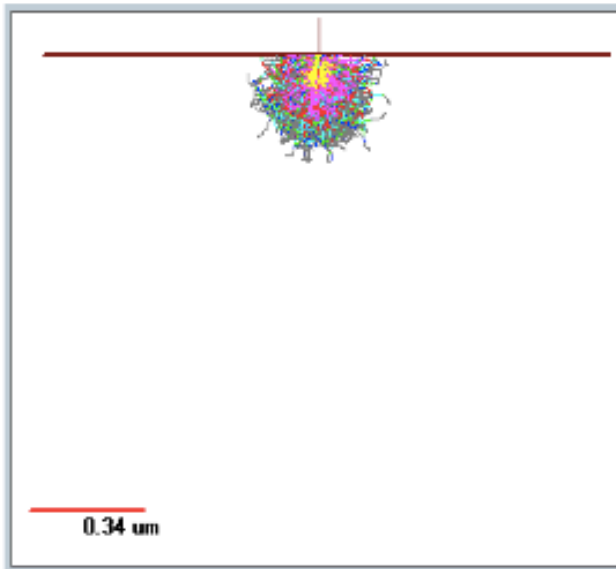




2015CE SIMULATIONS

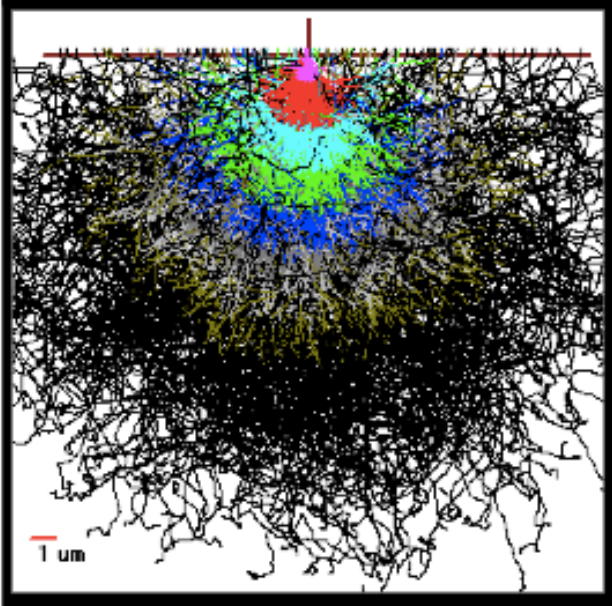


$\leftarrow e^-$   $He^+ \rightarrow$   
**10keV Si**

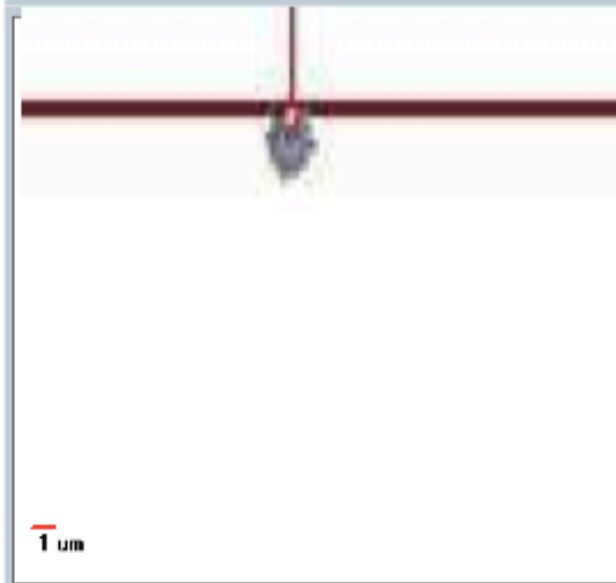


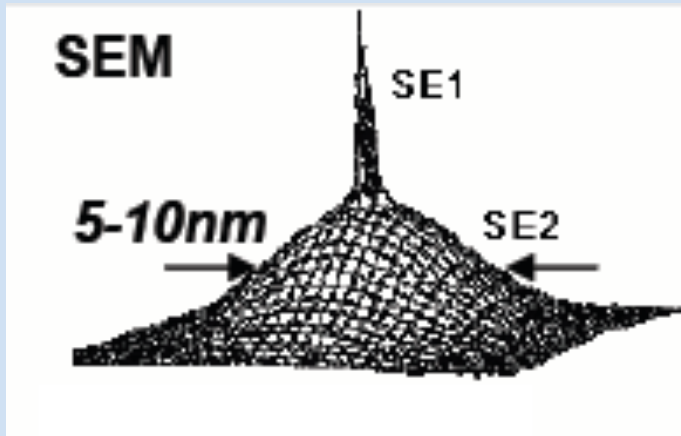
*Electron and ion trajectories plotted on the same size scale*

*Range Ratio @E  
(Electron/Ion) ~ E*

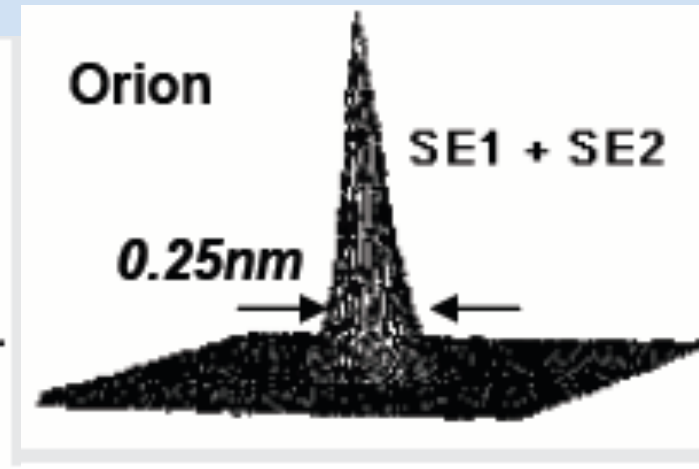


$\leftarrow e^-$   $He^+ \rightarrow$   
**30keV Si**





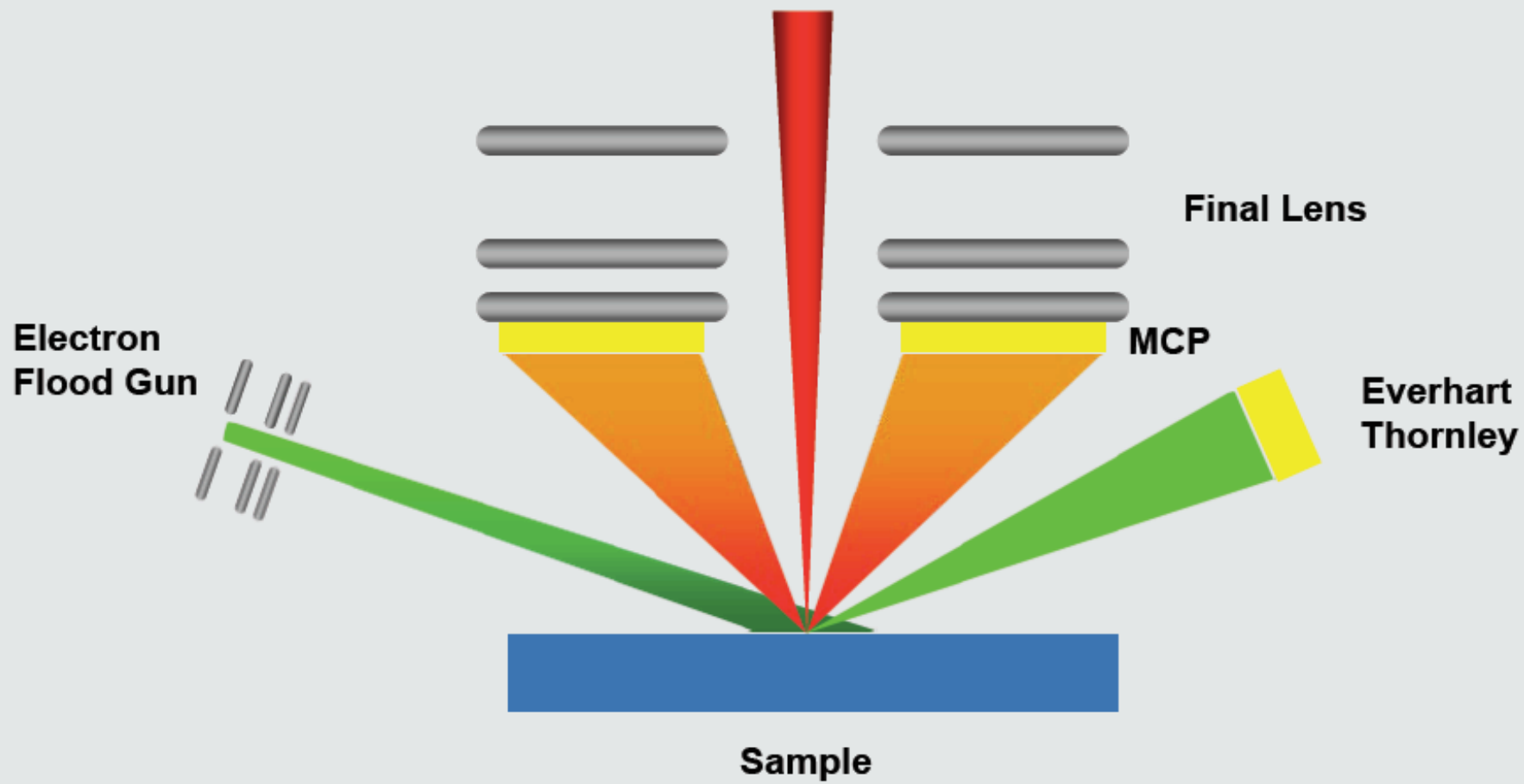
e beam induced secondary electrons



He ion induced secondary electrons

SE1 are produced directly by the primary beam

SE2 are produced by the backscattered beam as it exits sample



# The Helium Ion Microscope

- Similar in concept to the SEM but different in crucial details..
- Uses a  $\text{He}^+$  (or other) ion instead of electrons



**ZEISS ORION**  
installed at Harvard Courtesy  
Dr. David Bell

## Secondary Electrons

### Surface Information

Yield varies with material (range ~2-8)  
SE image provides both topographic information and material contrast  
Typical Energy ~5 eV  
SE Yield Ranges from 3 to 8  
Escape Depth ~ nm

## Backscattered Ions

### Material Information

Scatter yield varies as  $Z^2$  of target  
Scatter energy varies with mass of target (similar to RBS)  
Typical Energy ~ several keV  
Typical Yield ~ 0.1% to 3%  
Minimal Topographic Information  
Immune to surface charging

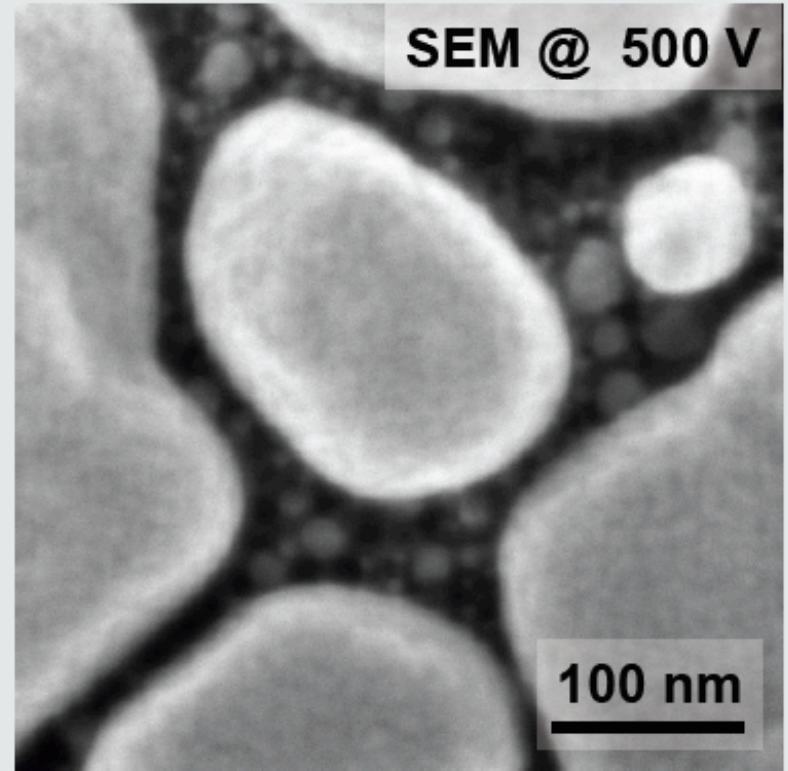
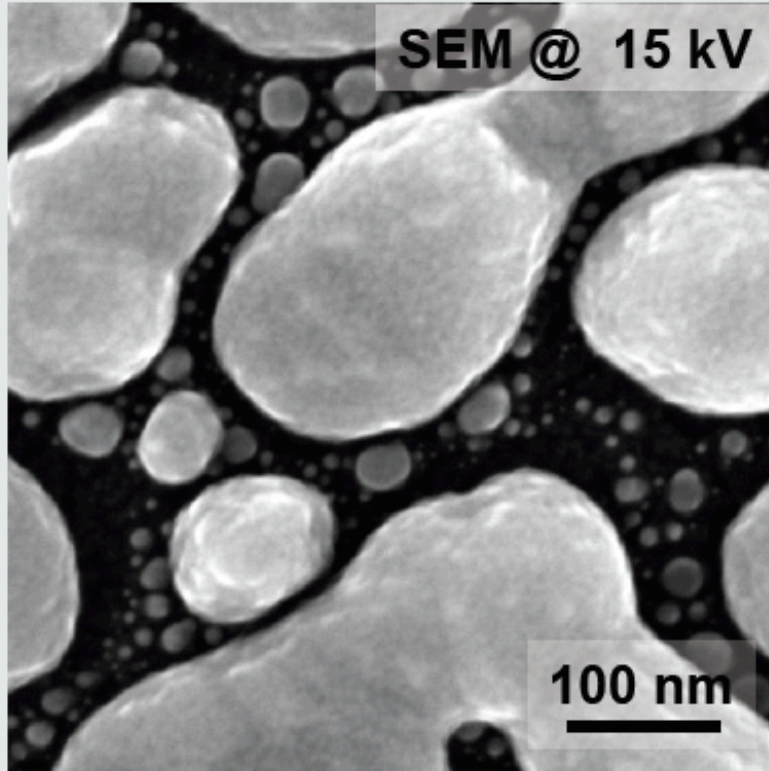
## Transmitted Ions

Provides material information.  
Scatter cross section varies with  $\sim Z^2$  of the target  
Provide crystallographic information

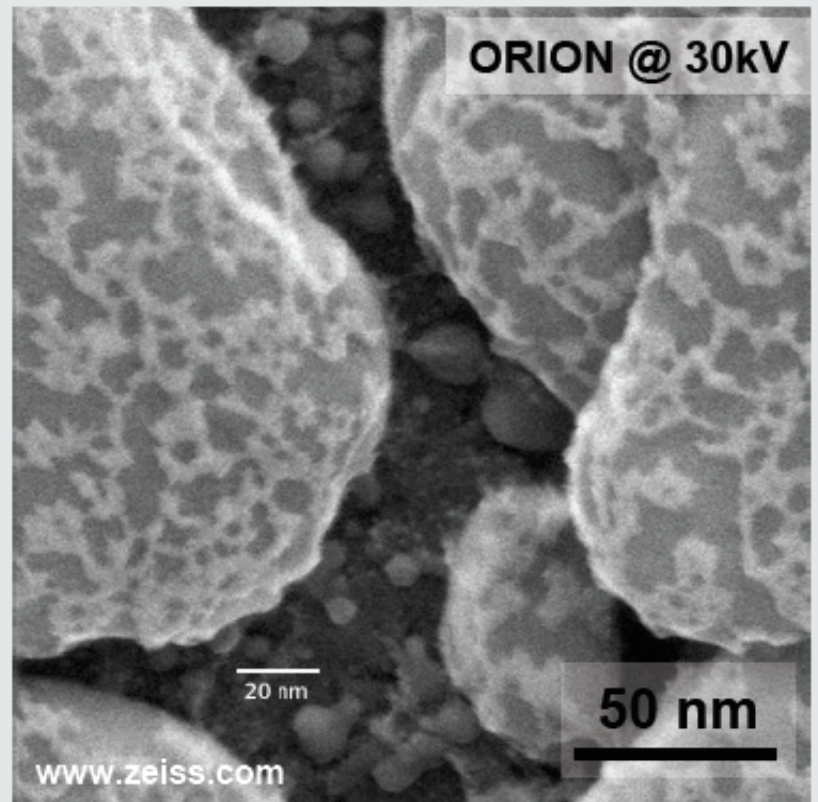
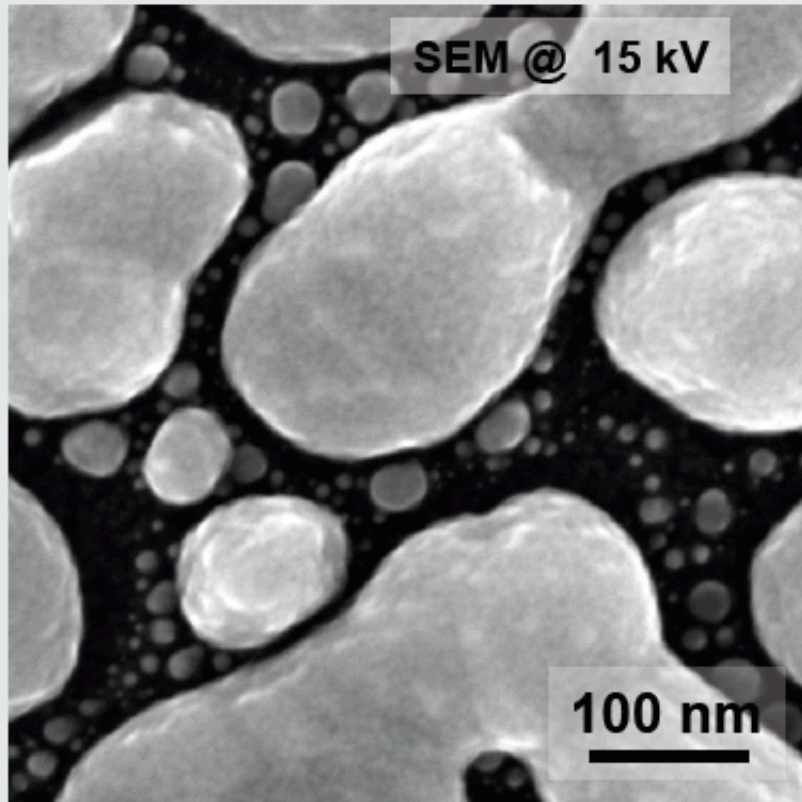
## Other Contrast Mechanisms

Channeling Contrast  
Voltage Contrast  
Photon Imaging “Ion luminescence”





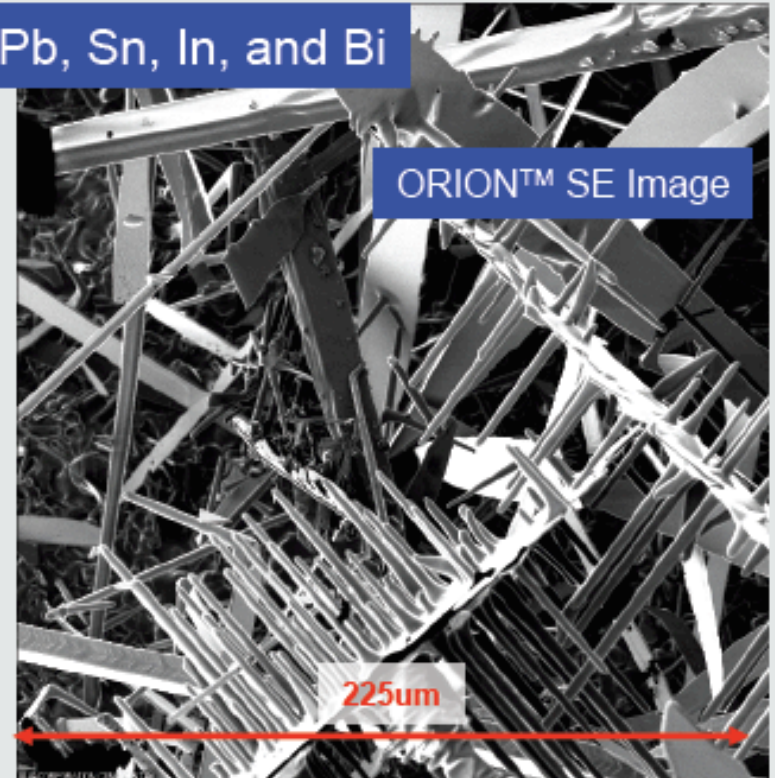
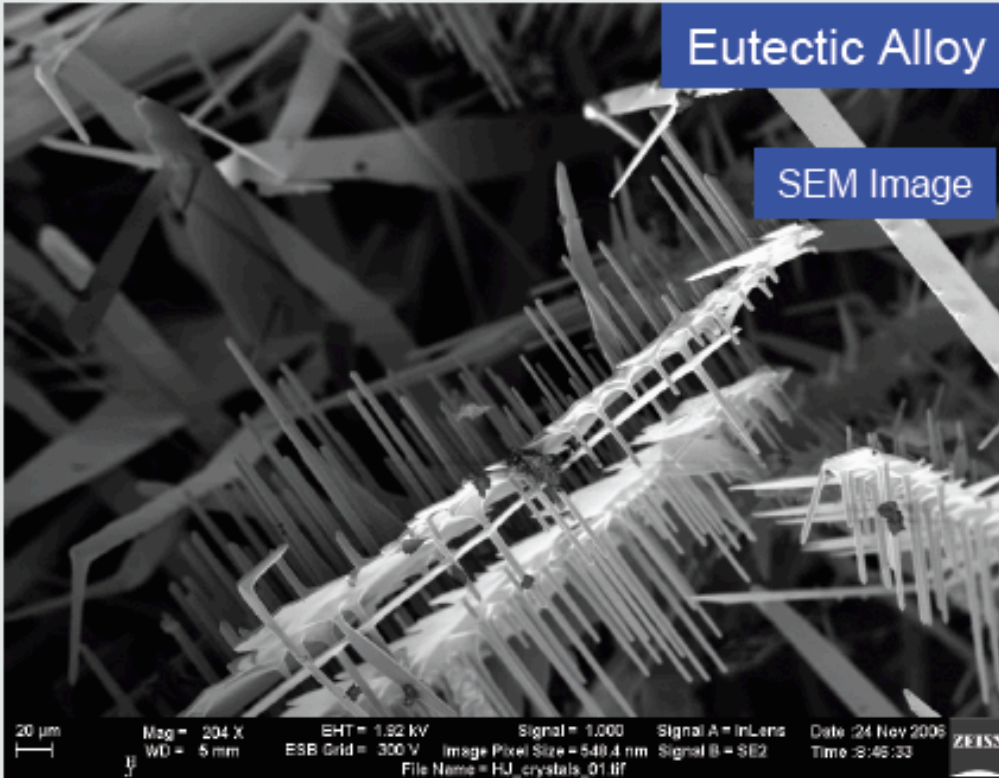
sample courtesy of Al Lysse, Carl Zeiss SMT Inc., US



## Eutectic Alloy of Pb, Sn, In, and Bi

SEM Image

ORION™ SE Image



Depth of Field inversely proportional to Half Angle ( $\alpha_i$ ) of incident beam



- contacted nanowire
- device bonded to its chip carrier
- WD = 9.5 mm !



20

This SEM image shows a top-down view of a device with a central nanowire. The nanowire is connected to a larger structure on the left. The image is rotated 90 degrees counter-clockwise. A scale bar labeled '20' is in the bottom right corner.



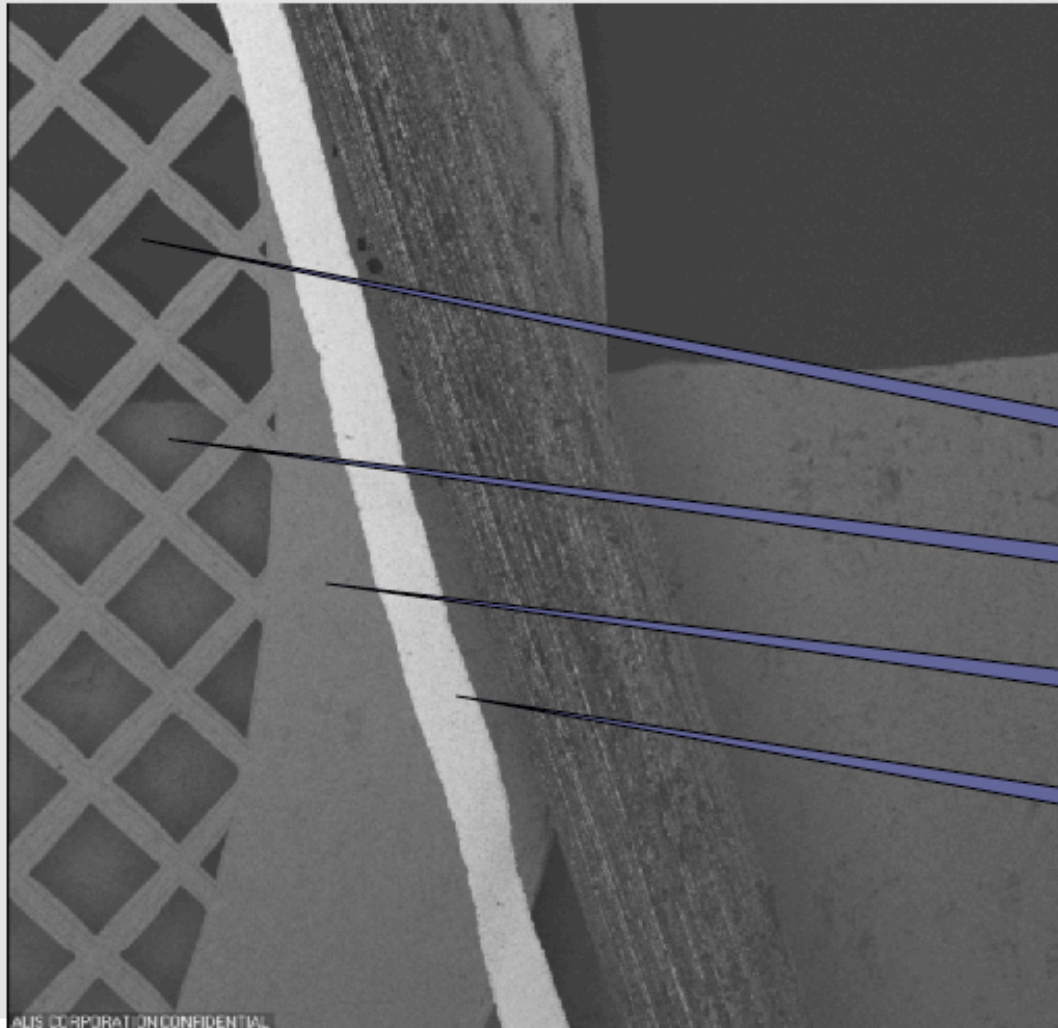
20

This SEM image shows a side view of a contacted nanowire device. The nanowire is connected to a larger structure on the left. The image is rotated 90 degrees counter-clockwise. A scale bar labeled '20' is in the bottom right corner.

- best imaging WD @ 6-8 mm
- advantage for imaging samples with big topography if ROI is not highest (e.g. curved samples)

courtesy of Prof. Jacob,  
EMPA Dübendorf, CH

Note how the RBI image contains very little edge information

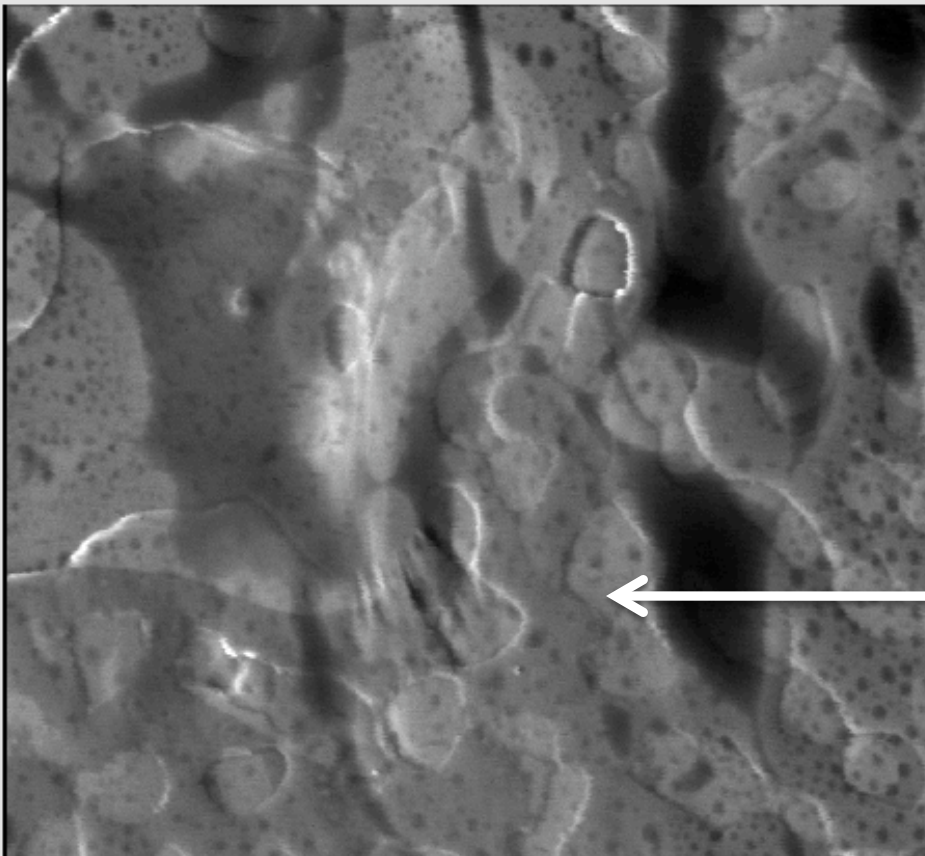


Carbon

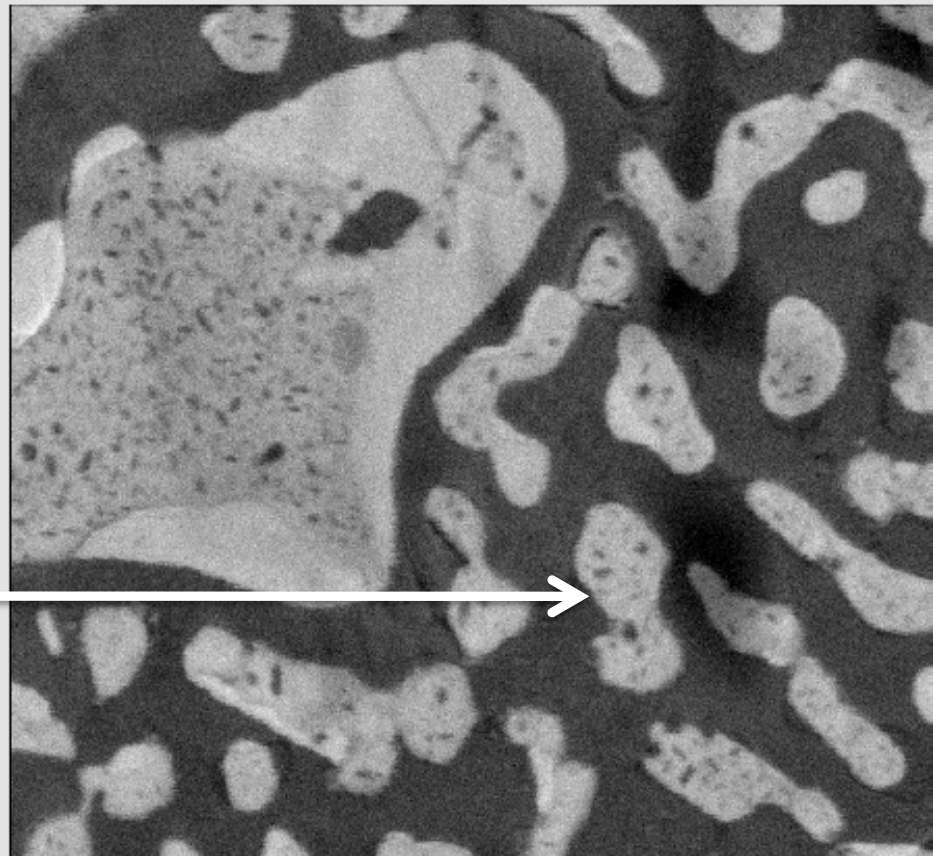
Nickel

Copper

Gold



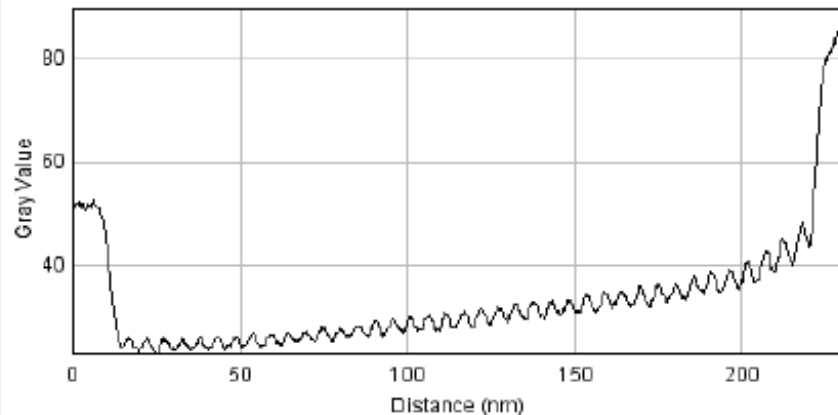
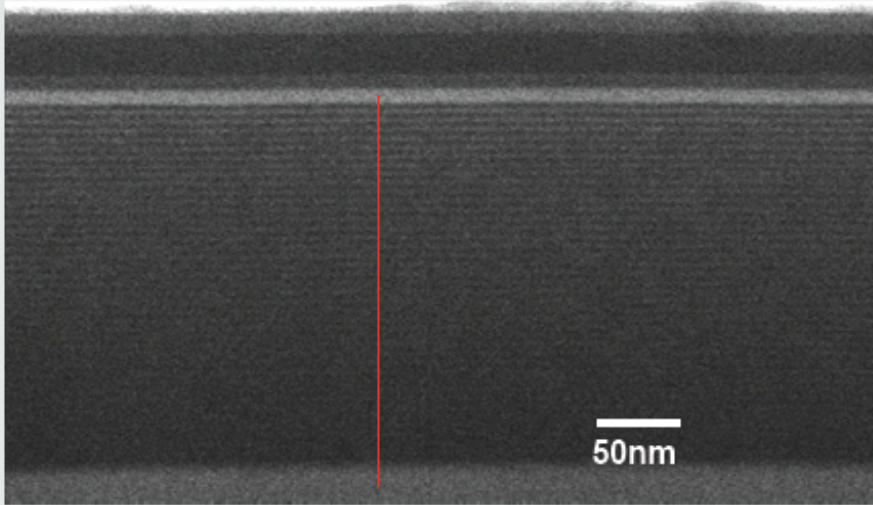
SE image



He Ion Backscattered image

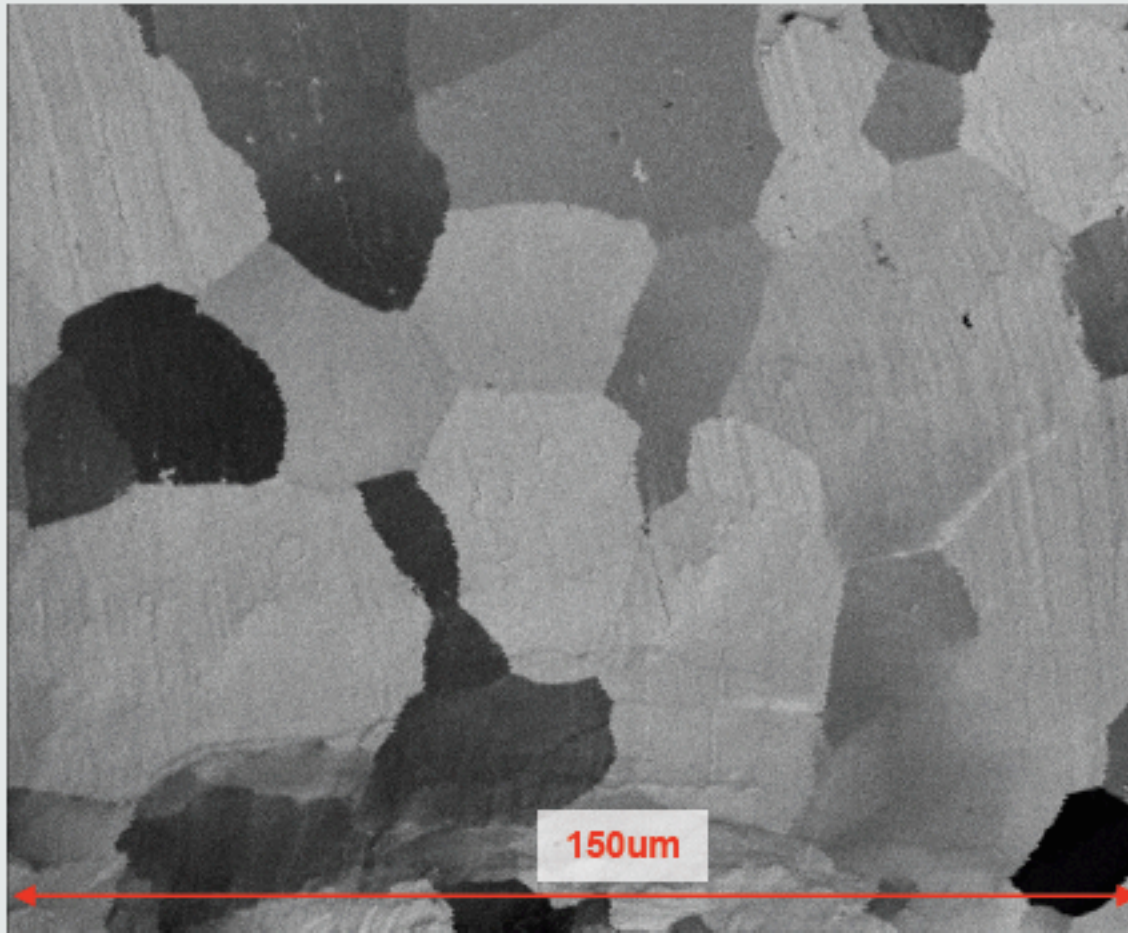




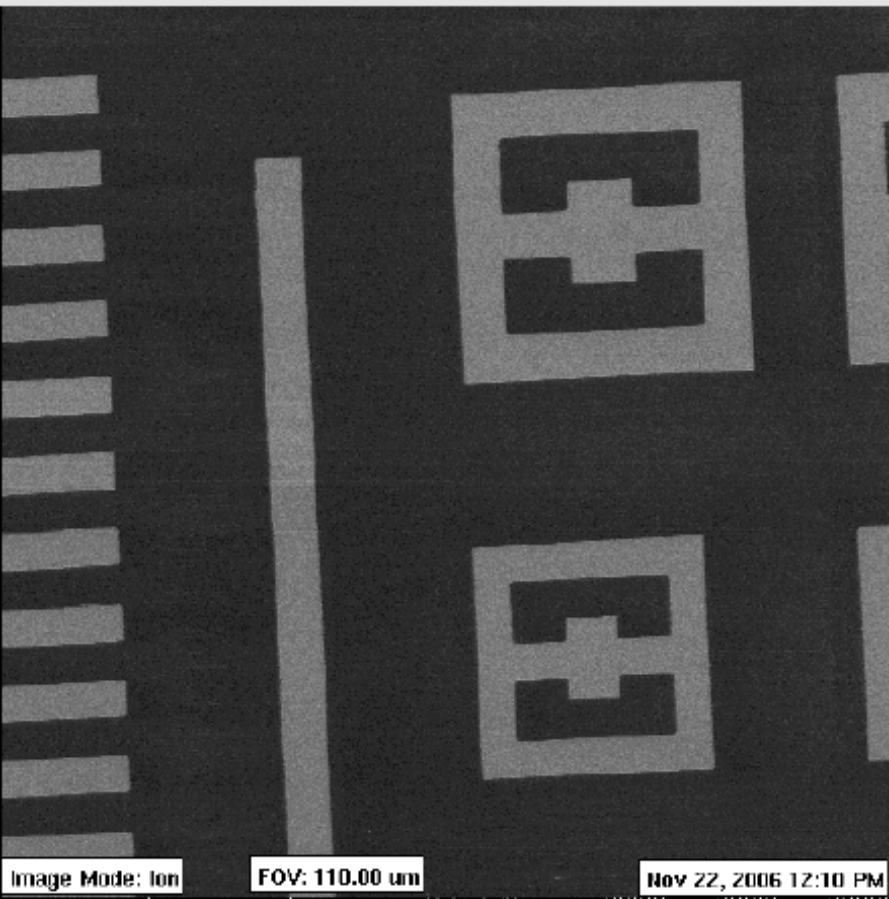


- Transmission Ion imaging performed with a "Conversion Style" detector
- Transmission image of EUV mask clearly shows the absorber stack, the cap layer, and the multi-layers
- An improved detector in development which will allow both bright and dark field imaging

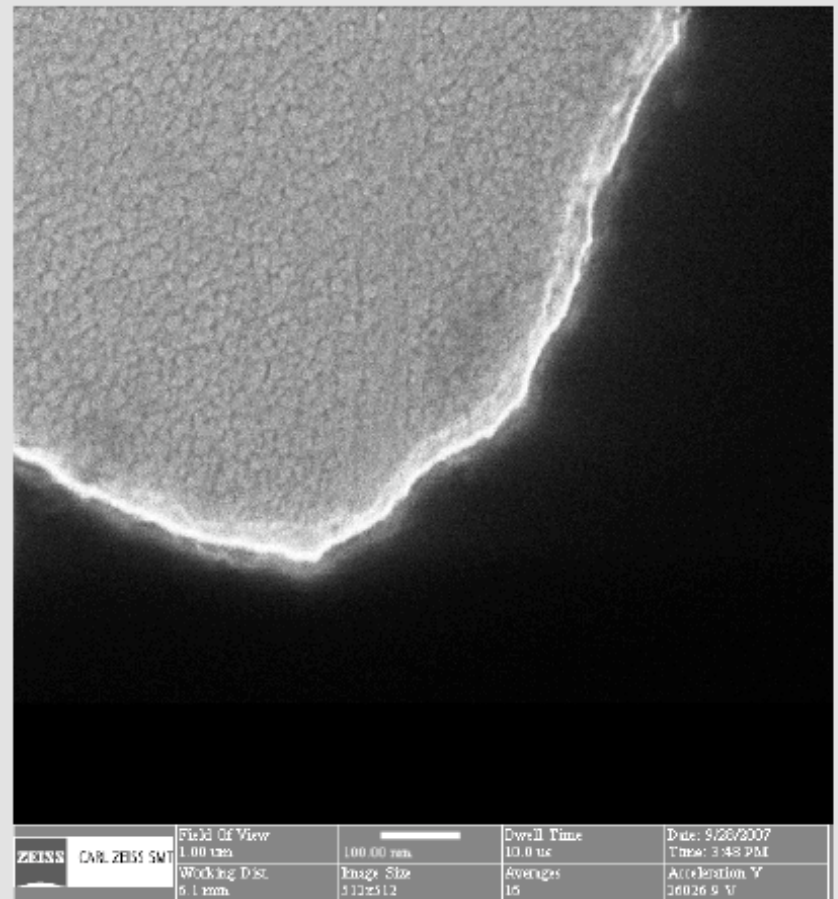
# Channeling Contrast



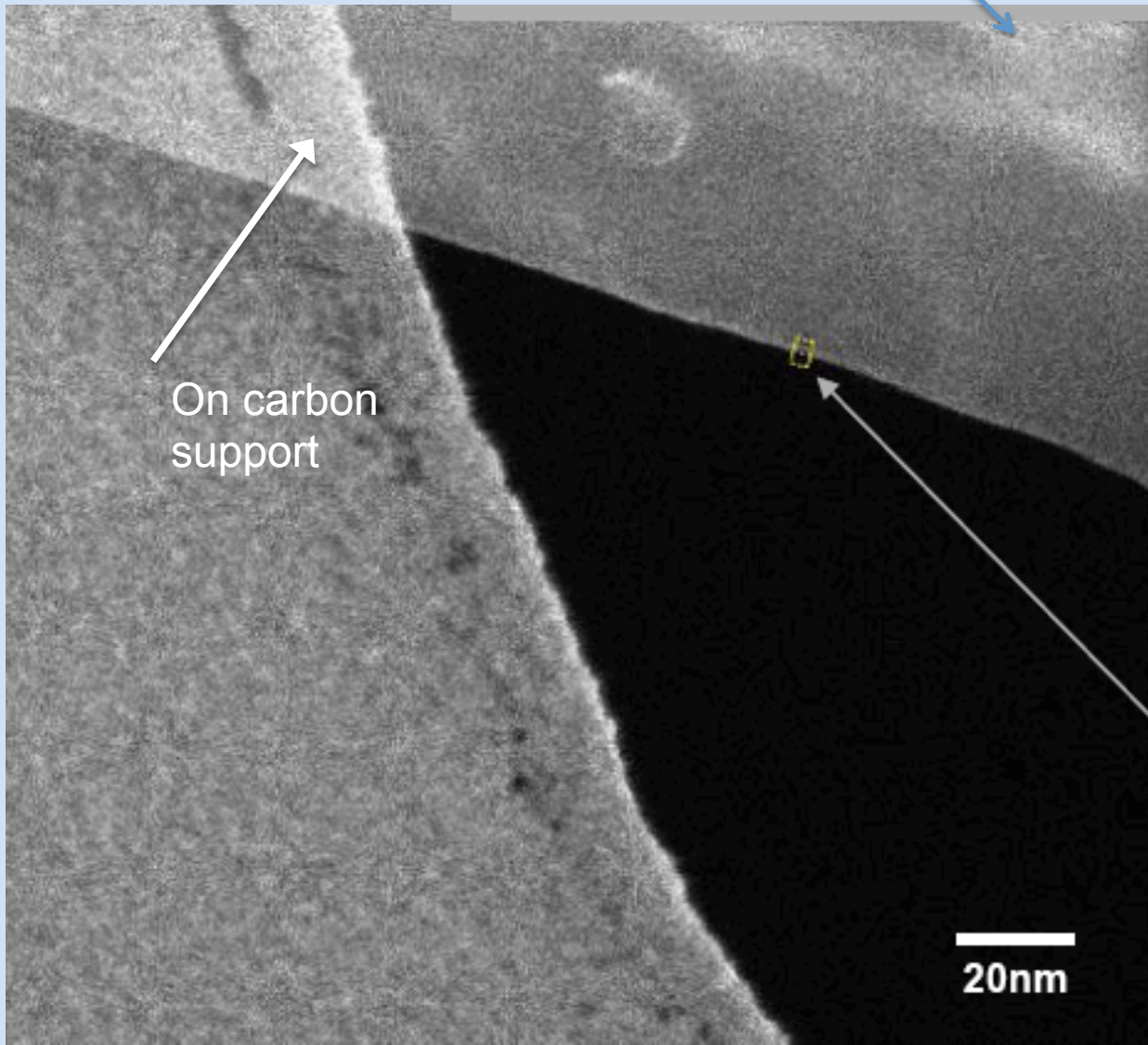
Contrast differences due to grain orientation



Backscattered ions have high energy and are impervious to surface charging



Charge control in Secondary electron mode using an electron flood gun



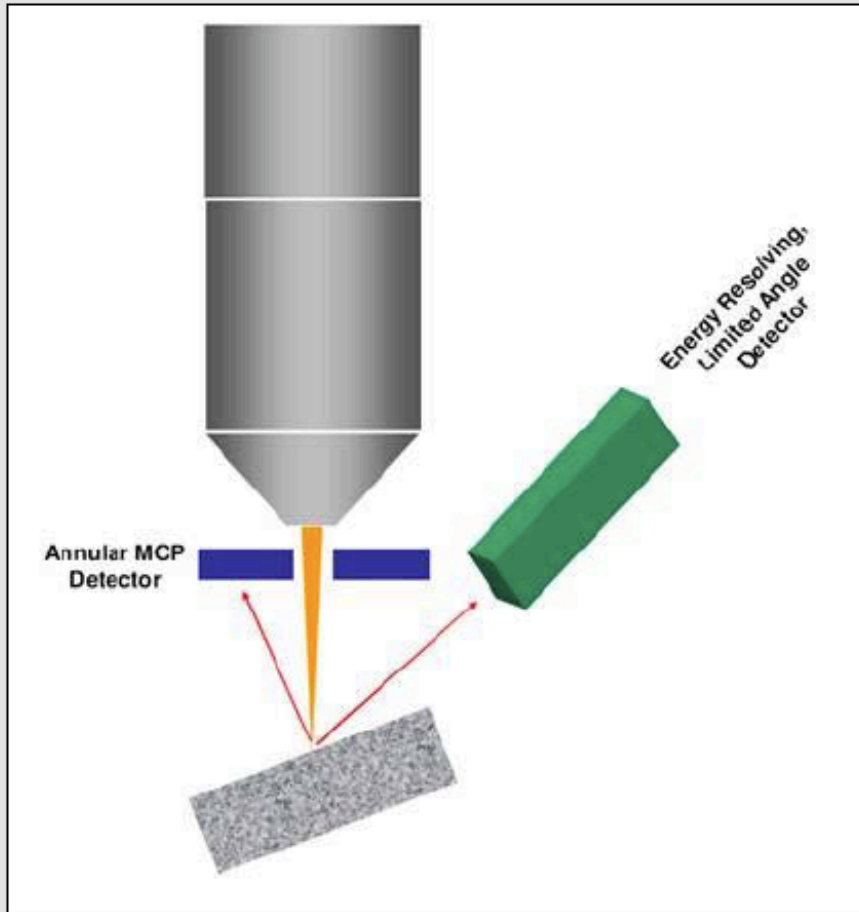
On carbon support

Self supporting

20nm

Asbestos Fiber on Holey Carbon Film





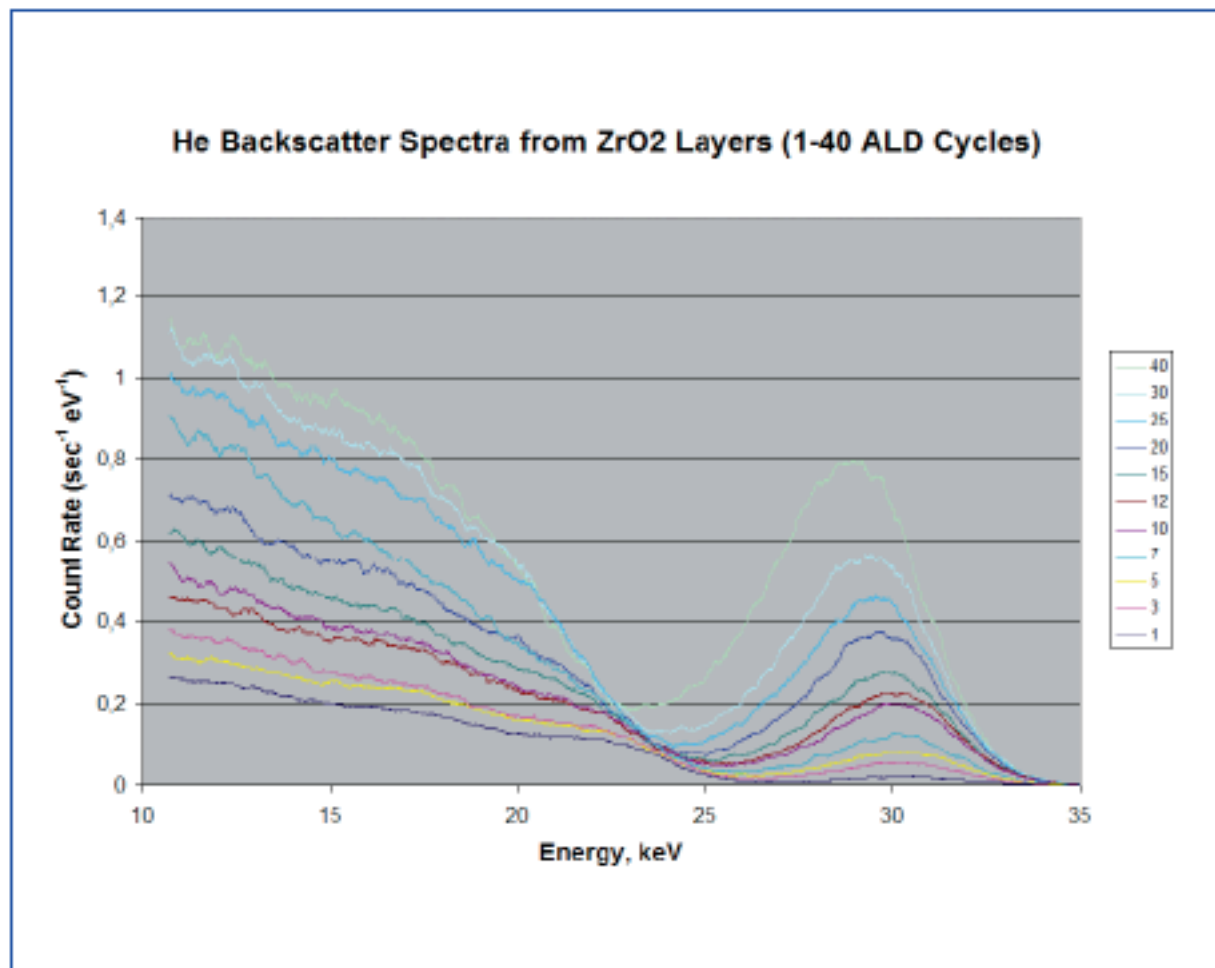
- Virtually no X-ray production → use backscattered helium for materials characterization
- Cross section for scattering (RBI Yield, Imaging)

$$\sigma \propto \frac{Z^2}{E_0^2} \frac{1}{\sin^4(\theta)}$$

- Backscatter Energy for spectroscopy

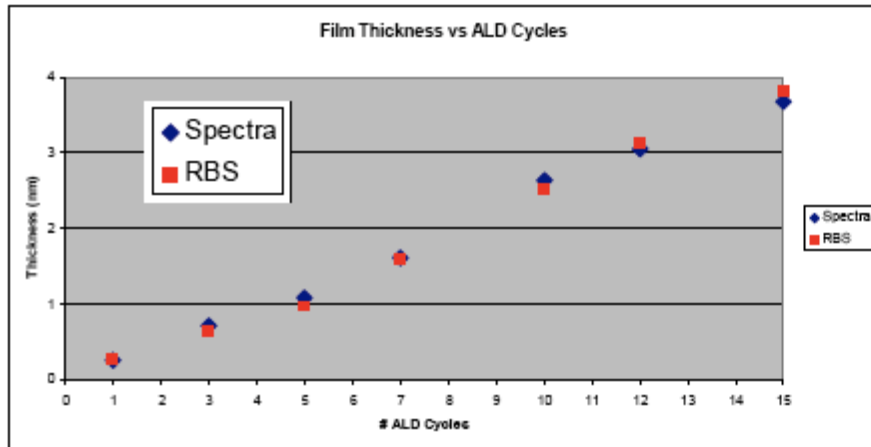
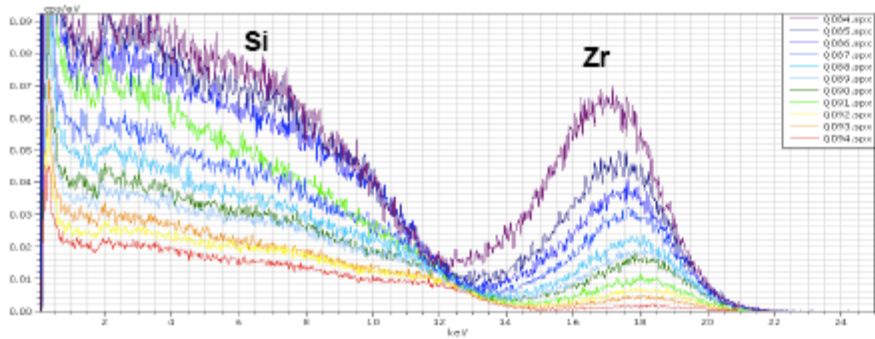
$$E_B = E_0 \left( \frac{\sqrt{M_2^2 - M_1^2 \sin^2 \theta} + M_1 \cos \theta}{M_1 + M_2} \right)^2$$





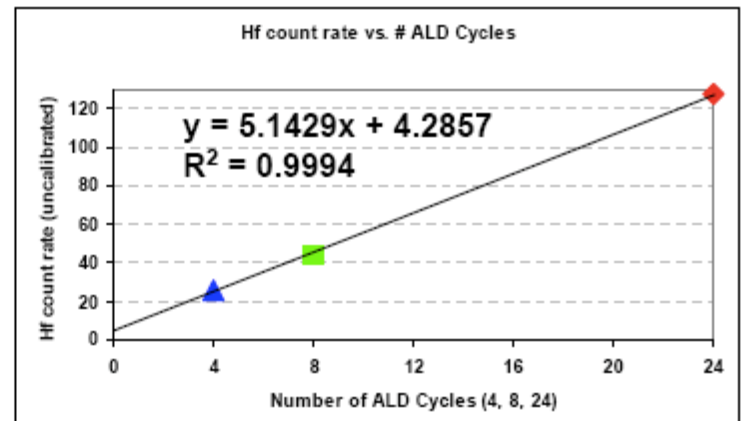
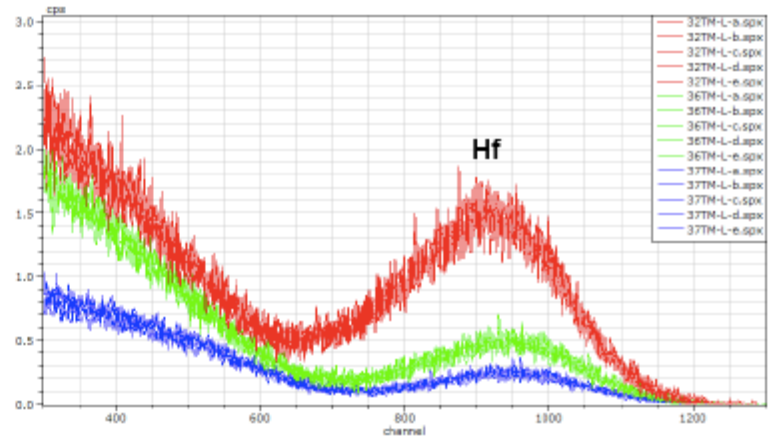
*Figure 1: Helium backscatter spectra from a set of ZrO<sub>x</sub> films on silicon. The legend indicates the number of ALD cycles used to grow the various films. Samples provided courtesy of Dr. Steffen Teichert, Qimonda (Dresden, Germany).*

## ZrO<sub>2</sub> on SiO<sub>2</sub>



Correlation between low voltage RBS and MeV RBS

## HfO<sub>2</sub> on SiO<sub>2</sub>



From B. Thompson, Zeiss

- The Helium ion microscope is based upon a gas field ionization source.
- Unique properties :
  - high resolution
  - different contrast mechanisms ( SE and RBI mode)
  - large depth of focus
  - high surface sensitivity
  - strong channeling contrast.
- The helium ion microscope can be used for fabrication.
- Disadvantages:
  - Source stability
  - Beam induced damage
  - Not fully developed technology

## References for Helium Ion Microscopy

B.W. Ward, J.A. Notte and N.P.Economou.  
JVST.B. 24. 2871 (2006).

D.C. Bell. Microscopy and Microanalysis.  
15.147-153 (2009)

G. Hlawacek, et.al. (2014). arXiv.  
11311.1711v2(cond-mat.mtrl-sci)